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Chapter 10

Challenges Associated with CO₂ Sequestration and Hydrocarbon Recovery

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Abstract

In the near- and midterm future, carbon capture and storage (CCS), also called CO₂ geo-sequestration, is likely to play a significant role in the reduction of atmospheric greenhouse gas. By expanding the set of possible sequestration targets, it is expected that CCS will enable larger quantities of CO₂ to be sequestered, mitigating human activity-driven climate change. In general, oil and gas reservoirs are ideal geologic storage sites for CO₂ because they have successfully held hydrocarbon molecules for millions of years. In addition to the significant and reliable storage capacity of hydrocarbon reservoirs, there is a considerable body of knowledge related to the behavior of hydrocarbon bearing reservoirs, and significant amounts of data are often acquired during their exploitation, factors which improve the economics and safety of any CCS project. By making use of existing and future oil and gas projects, CCS can become a major contributor in the fight against global warming, as well as a sizeable contributor to energy production worldwide. The CCS sequestration targets discussed in this study are sandstones, coal beds, shales, and carbonates. The potential and challenges associated with each of them are discussed in detail, and suggested topics for future research work are provided.

Keywords: CO₂ EOR, CO₂ storage, sandstone, carbonate, shale, coalbed methane

1. Introduction

Global levels of CO₂ in the atmosphere have been steadily rising with the increase of hydrocarbon production and usage. It is estimated that CO₂ emissions in the United States were approximately 5.5 billion tons in 2015, the largest volume yet. Anthropogenic greenhouse
gases, such as carbon dioxide (CO₂), are considered a major contributor to global warming [1]. Sequestration of power plant-generated CO₂ through injection into petroleum and gas reservoirs through a process called carbon capture and storage (CCS) or “carbon sequestration” has been proposed as a method for reducing greenhouse gas emissions. Research on the use of CO₂ for enhanced oil recovery (EOR) continues with growing interest; however, research concerning terrestrial sequestration of CO₂ for environmental purposes, such as CCS, is relatively recent. As a result, fundamental topics of interest in sequestration research are concerned with scientific and technical aspects, as well as practical concerns such as the economic feasibility, safety, and the maximum possible amount of CO₂ storage [1]. Therefore, fighting CO₂ emissions with EOR and CCS is a priority, leading to innovations within the petroleum industry.

The process of CCS involves pumping sizeable quantities of atmospheric CO₂ underground, where, under the right circumstances, it can remain safely sequestered for thousands or millions of years. The economics of CCS are often unfavorable, especially as CO₂ is generally an expense rather than a revenue stream, but by combining the end goal of CCS with enhanced oil recovery (EOR) techniques used in the oil industry, there is the potential that CCS can be made economical while also increasing the productivity and efficiency of existing oil resources.

CO₂ EOR generally involves the injection of CO₂ into an oil-bearing reservoir in order to decrease oil viscosity, decrease the interfacial tension between oil and water, and increase the elastic energy of the formation, generally resulting in improved oil production. In the case of methane-bearing formations, most notably coal beds, injected CO₂ has a far stronger affinity to the formation than methane, resulting in the replacement of adsorbed methane with adsorbed CO₂ both increasing methane production and resulting in the sequestering of large volumes of CO₂.

EOR and CCS projects are both complicated tasks that require a vast understanding of the target reservoir in order to enhance storage capacity and storage time of CO₂, as well as hydrocarbon production. These topics will be discussed in greater detail throughout this paper.

1.1. Trapping mechanisms

One of the primary considerations when approaching a CCS project is the different mechanisms by which CO₂ can become safely sequestered underground. Generally, there are four different trapping mechanisms employed in the sequestration of CO₂, each of which contributes differently to the duration and volume of CO₂ trapping (Figure 1). In the different time stage, those four trapping mechanisms will work together.

- **Structural/stratigraphic trapping:** These types of traps are formed from tectonic forces and generally involve physical barriers to flow. An example of this is a thick layer of low permeability rock (caprock), such as shale, where, assuming a favorable structure, rising CO₂ will become trapped and begin accumulating.

- **Residual trapping:** This phase of trapping starts as soon as the CO₂ is injected. While the CO₂ is being injected, it is displacing the fluids that are inside the pores of the formation. As the primary CO₂ volume migrates upward, small volumes of CO₂ remain inside these tiny
pores due to capillary forces. This mechanism immobilizes the CO$_2$ potentially storing it in the formation for millions of years, just like the fluids it displaced.

- **Solubility trapping:** Solubility trapping refers to CO$_2$ being absorbed or adsorbed within the formation. Absorption occurs as CO$_2$ dissolves in the formation fluid, while adsorption occurs as CO$_2$ binds to the surface of the formation, like a piece of metal attaching to a magnet. After CO$_2$ has been adsorbed in the fluid, it will exist as a mixture, which will not be as buoyant as its gaseous form and will not migrate upward through the formation. This mixture will be denser than the surrounding fluids and will migrate downward over time.

- **Mineral trapping:** At the time of injection, this type of trapping is insignificant. Over a long period, after CO$_2$ has been dissolved in the formation fluids, it will begin reacting with the minerals in the surrounding formation and create solid carbonate minerals. These solid carbonate minerals will be attached to the rocks it reacted with and can be stored in the formation for millions of years.

Overall, these trapping mechanisms prevent carbon dioxide’s upward travel and leakage while increasing the CO$_2$ storage potential and security of the desired formation. Assuming an ideal trapping mechanism, the temperature-related properties of a reservoir must be considered as well. The required temperature to store CO$_2$ underground should be less than the critical temperature of CO$_2$, making reservoirs such as those in Illinois Basin prime candidates. The critical temperature of CO$_2$ is 87.7°F; naturally, most geological formations exceed this temperature due to geothermal gradient [2].

### 1.2. Sandstone reservoirs

Sandstone reservoirs were the primary source of oil production during the early life of the oil industry. Many wells were produced and then abandoned long before the introduction of enhanced oil recovery (EOR) and other modern techniques that have enabled production
from formations once thought of as nothing more than barriers to flow or geologic curiosi-
ties. In this modern day, sandstone reservoirs, once the workhorse of the oil industry but
long since abandoned due to declining production, can be made to once again flow in eco-
nomic quantities through the use of EOR techniques, such as CO₂ injection. While ensuring a
renewed flow of oil to an energy-hungry world, CO₂ EOR in these old sandstone reservoirs
may also play a major role in the preservation of our environment as injected CO₂ can be
sequestered in subsurface formations for thousands of years. With these unique opportunities
come unique challenges, ranging from the significant reservoir analysis required to ensure a
safe sequestration to the infrastructure required to deliver such sizeable quantities of CO₂.

Sandstone reservoirs are particularly notable due to the sheer number of wells drilled in such
formations that have been produced throughout the history of the oil industry and have since
been abandoned. Due to their number, as well as how much time we have had to accumulate
knowledge about their behaviors and the petrophysics involved in their production, sand-
stone reservoirs are likely to play a major role in any large-scale CCS program.

1.3. Coalbed reservoirs

Another ideal medium in which to store CO₂ is coal beds. Generally used to produce coalbed
methane or coal at shallower depths, coal beds have a dual porosity system, which can be
classified as primary and secondary porosity system. The pores within the coal matrix make
up the primary porosity, while the pore volume of the numerous fractures permeating the
coal bed makes up the secondary porosity.

The methane that is the primary target of coalbed drilling is stored in the coal matrix via
adsorption. Because CO₂ has a greater affinity for coal than methane does, CO₂ is the desired
choice to enhance methane recovery, and coal beds are a good place to store CO₂. Coal beds
are distinctively different from the conventional hydrocarbon reservoirs in production as
well as gas storage mechanisms. In conventional oil reservoirs, CO₂ is dissolved in the oil to
decrease the viscosity of the oil, resulting in a great deal of CO₂ being recovered at the surface
along with the produced oil. The “sequestered” CO₂ is then only that which dissolves into
residual oil or is trapped due to one of the other trapping mechanisms [3]. In the case of coal
beds, the majority of CO₂ adsorbs directly to the surface of the coal bed, providing a more effi-
cient mechanism of sequestration while also forcing methane off the coalbed surface, helping
to release any residual gas production. For example, methane recovery was improved from
77 to 95% of original gas in place at the Allison Unit CO₂-ECBM pilot in the San Juan Basin
[4]. Coal beds can act as a significant contributor to CCS through the excellent economics of
CO₂-based EOR, as well as the quality of their sequestration.

1.4. Shale reservoirs

As technology has advanced throughout the years, oil and gas exploration in unconventional
shale reservoirs has become the main focus of the oil industry. Horizontal drilling and the
hydraulic fracturing of shale formations have allowed us to unlock vast reserves of oil and gas
production. Considering shale formations have extremely low permeability (of nanoDarcy in
some cases), primary production does not produce the maximum amount of oil possible out
of the formation. In most cases, tertiary production or EOR will begin with gas injection, such as carbon dioxide (CO$_2$), instead of water flooding because of shale’s low permeability and the risk of reactions between the clays and injected water. During this production enhancement, some of the injected carbon dioxide will be permanently stored in the formation with different storage mechanisms, while some will be produced with the oil stream and get recycled back into the formation. CCS in shale reservoirs is often more difficult as less is known about their geology and long-term behaviors.

Shale reservoirs will likely play an important part in future CCS projects due to the scale of many shale reservoirs, their quality as a seal, and the importance of EOR techniques in existing shale plays.

1.5. Carbonate reservoirs

CO$_2$ injection into carbonate reservoirs was first considered in the 1930s but did not become a reality until 1964 in the Mead Strawn field located in Texas. CO$_2$ injection has since been established as a reliable form of EOR, with results regularly matching or surpassing those of other EOR techniques. In the 1964 example with the Mead Strawn field, oil production was increased by up to 82% beyond the results of a standard water flood [5]. Like many sandstone reservoirs, carbonate reservoirs have a long history and will likely play a significant role in future CO$_2$ EOR and CCS projects.

Hill et al. [6] state carbonate CO$_2$ EOR now produces approximately 305,000 bbls worldwide with an accelerating growth rate. The areas targeted for carbonate CO$_2$ projects in the United States are as follows: Louisiana, Michigan, Mississippi, New Mexico, Oklahoma, Texas, and Wyoming. CO$_2$ production wells provide immense amounts of data on the reservoir response to a CO$_2$ flood compared to saline projects. Azzolina et al. [7] discuss how CO$_2$ EOR is an established method for extending the life of a hydrocarbon sustaining carbonate reservoir.

Dissolved CO$_2$ injection into carbonate subsurface formation increases geologic carbon storage integrity by avoiding dependence on trapping mechanisms. As a result, solubility trapping will dominate until mineral trapping occurs, which is dependent on the formation rock [56]. Izgec et al. observed that solubility storage of CO$_2$ is larger than mineral trapping [1]. Eke et al. [8] state geological CO$_2$ storage in carbonate formations for long timescales (sequestration) relies on the contribution of several CO$_2$ trapping mechanisms: physical trapping in a subsurface formation, solubility trapping, hydrodynamic trapping, and mineral trapping.

2. Existing field applications

With the need for the prompt reduction in CO$_2$ emissions, the development of CCS must be taken seriously, as it has the potential to make a major difference in the levels of atmospheric CO$_2$. At one time, it was believed that oilfield reservoirs did not have sufficient pore volumes to have a significant impact on CO$_2$ emissions, but it is now understood that not only are there massive pore volumes available for CCS in depleted major pay zones (MPZs) of reservoirs,
but there also exist residual oil zones (ROZs) and transition zones (TZs) in hydrocarbon fields that can be depleted and used for sequestration through quaternary production.

Traditionally, residual oil zones (ROZs) are considered to be uneconomic by the end of their primary or secondary recovery phase due to their extremely low oil saturation. However, Advanced Resources International [9] analyzed the feasibility of using CO₂ EOR to recover hydrocarbons from the ROZ and determined that a total of 55 fields in the Permian Basin have the potential to become economic ROZ resources. Simulations using CO₂ PROPHET, a water and CO₂ flood prediction software available through the US Department of Energy (DOE) website, estimated the recoverable ROZ at 11.9 billion bbls of the 30.7 billion bbls of TZ/ROZ oil in place in these five Permian Basin oil plays [9].

Usage of CO₂ injection as a form of EOR has not been limited to pilot and research tests. Kinder Morgan estimated that in the past 37 years, 655 million tons of CO₂ have been injected, produced, and recycled back into EOR. This is an average of 17.7 million tons per year, which is enough to negate the yearly emissions of six 500 MW coal-fired electric power plants [10]. Examples of some of these different field applications are given below.

2.1. Permian Basin

The Permian Basin in West Texas is one of the largest areas employing CCS techniques in ROZ and TZ and is currently undergoing the largest CO₂-enhanced oil recovery (EOR) operation in the world. Most of the ROZs created in this area are due to lateral sweep by hydrodynamics and have thicknesses in excess of 300 feet [11]. While implementation of CO₂ floods is not particularly widespread due to the limited availability of CO₂, the Permian Basin has ready access to a pipeline of CO₂ originating in natural supplies in Colorado and New Mexico. Of the six CO₂ EOR projects in which recovery response has been published for Gulf Coast sandstone reservoirs, recovery factors are from 15 to 23% of original oil in place (OOIP) [12].

2.2. Port Neches

A CO₂ injection project in Port Neches, in a Texas Gulf sandstone, started in September 1993. The field had previously undergone water flooding, leaving a residual oil saturation of 30%. The goal of the project was to recover an additional 10% original oil in place (OOIP) [13]. A follow-up paper recorded that the production peaked at 500 barrels of oil per day (Bopd) (Figure 2) and later at 800 Bopd with CO₂ injection. The OOIP reduced from 12 to 7 million stock tank barrels (MMSTB) in the main fault block of the reservoir [14].

2.3. Bati Raman field

In 1986, the Turkish Petroleum Corporation started a large immiscible CO₂ injection project; the trend can be seen in Figure 3.

2.4. Ordos’ Basin

The evaluation of Changqing oil field, Ordos’ Basin, Northwest China, concluded that conducting a CO₂ flood after water flooding could produce 119 million tons of oil and sequestrate 273
million tons of CO$_2$ [16]. In 2000, the International Energy Agency Weyburn CO$_2$ Monitoring and Storage Project did a study on CO$_2$ storage in a partially depleted oil reservoir and found that a $1.5$ billion, 30-year commercial CO$_2$ EOR produced an additional $130$ million barrels of oil.

![Figure 2. Production vs. time plot.](image)

![Figure 3. The Bati Raman field's production trend [15].](image)

### 2.5. SECARB

The Southeast Regional Carbon Sequestration Partnership (SECARB) [17] operated a test for CO$_2$ sequestration at Black Warrior Basin in Alabama from 2006 to 2009 and determined that more than $360$ million tons could be sequestered while increasing coalbed methane reserves by more than $20\%$. The SECARB set up monitoring systems in shallow boreholes and continues to monitor the local soil profile to determine if seepages of their test injection of $1000$ tons of CO$_2$-injected gas occur and to facilitate the development of monitoring protocols that will ensure the safe conduct of CO$_2$ injection activities.
2.6. SWP projects

The Southwest Regional Partnership for Carbon Sequestration (SWP) indicates that over 2 million metric tons out of a total of 7 million metric tons retained CO$_2$ in the Scurry Area Canyon Reef Operators (SACROC) project were dissolved in the aqueous phase. That report does not include nor report CO$_2$ dissolution in oil, and therefore the numbers for CO$_2$ dissolution in the aqueous phase may be compromised. In addition to CO$_2$ dissolution in oil, the presence of a hydrocarbon phase can limit the contact between injected CO$_2$ and the aqueous phase even in depleted carbonate reservoirs. This work will, therefore, enhance estimates of predicted storage capacity both in depleted and producing oil reservoirs by revisiting and considering CO$_2$ solubility in the oil phase.

Additionally, rock wettability determines whether hydrolyzed CO$_2$ and the resulting acid in the aqueous phase can come into contact with the rock surface. When the rock is strongly oil wet such as in most carbonates, carbonate dissolution cannot take place; therefore, requirements for the mineralization trapping mechanism will not be met. In that case, the current estimation of CO$_2$ storage capacity in oil reservoirs because of the mineralization mechanism should be revisited. There is no indication of wettability measurement in the SACROC project. The SACROC project seeks to develop a subsurface geochemical-compositional flow model that incorporates the physics learned from lab-based measurements conducted throughout the course of its work, which will add considerably to the body of knowledge for carbonate reservoirs.

2.7. Existing exploited CO$_2$ sources

The majority of CO$_2$ injected into formation during operations is from natural reservoirs; however, problems arise such as climate change, diminished supply, and large demand. Innovation provides the solution by capturing CO$_2$ previously released to the atmosphere and using it for CO$_2$ EOR. During the production process, produced CO$_2$ is captured at the surface and reinjected, thus trapping the majority of injected CO$_2$ in formation. In Wyoming, natural gas processing plants produce approximately 716 Tcf of CO$_2$ while injecting 705 Tcf [7] in carbonate formations. In Michigan, an existing source of CO$_2$ provides the opportunity for carbonate CO$_2$ EOR in the NPRT; thus, Core Energy is using CO$_2$ emissions for EOR operations exploiting carbonate reef deposits [18]. These examples are helping reduce the emissions that would otherwise be vented to the atmosphere.

2.8. ECBM studies

Due to the effectiveness of CO$_2$ EOR and sequestration in coal beds, numerous studies have examined the usage of CO$_2$ sequestration in enhanced coalbed methane (ECBM) fields, and there are many field cases.

Mastalerz et al. [19] studied CO$_2$ sequestration and ECBM in unminable coal seams of the Illinois Basin. They found that approximately 271 billion tons of CO$_2$ could potentially be sequestered in the basin. Moreover, they found that potentially 1.6–4.6 billion tons of CO$_2$ could be sequestered in Illinois Basin coals and 70–280 billion m$^3$ (2.4–9.8 Tcf) of CH$_4$ is potentially recoverable as a result of CO$_2$ ECBM practices. The paper does suggest that
volumetric strain due and coal swelling, which causes permeability damage, should be considered in any CCS or CO$_2$ EOR project.

Yu et al. [20] predicted in 2007 that the CO$_2$ sequestration throughout all ECBM projects (existing and potential) in China could result in over 3.751 Tm$^3$ of additionally recoverable methane, with a CO$_2$ sequestration capacity of around 142.67 billion tons.

### 2.9. Shale storage capacity (the United States and Canada)

The amount of available storage for CO$_2$ in oil and gas shale is currently unknown, but the vast volumes of shale formations indicate that the storage capacity is significant. A recent report has estimated between 1.85 trillion and 20.5 trillion tons of carbon dioxide storage capacity is available in oil and gas reservoirs just in the United States and Canada. These estimates suggest the availability for storing centuries worth of CO$_2$.

### 2.10. Additional possible locations and projects

Depleted oil and gas fields in the SECARB region could provide 29.7–34.7 billion tons of CO$_2$ storage with 24 million recovered oil barrels [21]. Almost 60% of the estimated volume relate to offshore fields. Coal and organic-rich shale formations can also offer a significant place for storage due to high absorption capacity of CO$_2$, in addition to potential EOR applications. A tertiary coal in the Gulf of Mexico is estimated to have 20–28 billion tons of CO$_2$ storage [18].

The potential storage capacity of the Barnett Shale is estimated to be 19–27 Gton, while other shale formation, Fayetteville Shale, is estimated to be capable of sequestering 14–20 Gton of CO$_2$ [18]. There are still a lot more fields in the SECARB region to be evaluated on a possibility of a potential CO$_2$ storage and sequestration site. The SECARB region has a large annual CO$_2$ emission from coal-fired power generation and other fossil-fueled plants. In 2008 it was estimated to emit almost 2.9 [22] billion metric tons of CO$_2$.

An estimation of possible CO$_2$ sequestration volume was done by a “production replacement” principle, where for every volume of hydrocarbon, a 1:1 replacement ratio of CO$_2$ volume takes place. For the 2008 rates of CO$_2$ emission, SECARB region was capable of providing at least 28 years of CO$_2$ storage [18]. A case with a CO$_2$ EOR and sequestration in the Bell Creek oil field has a promising estimation of a recovery of additional 35 [23] million bbl of incremental oil through CO$_2$ flooding. Current plans exist to build a 232-mile pipeline from ConocoPhillips Lost Cabin gas producing plant to the Bell Creek field. This will help to integrate the large-scale storage of over 1 million tons of CO$_2$ per year.

### 3. Upcoming improvements to field applications

Kuuskraa, Godec and Dipeitro [24] analyzed primary and enabling next-generation technologies with applications in CO$_2$ sequestration, as shown in Table 1, and approximated the benefits of these technologies on a sample field area, as shown in Table 2. Notably, using their sample and estimates, they predict an increase in economically recoverable resource from 21.4 to 63.3 billion bbls.
Ahmadi et al. [25] performed numerical modeling to investigate reasonable CO\(_2\) injection scenarios, which were applied to CO\(_2\) sequestration and ECBM. In their study, the main goal was to study different CO\(_2\) injection methods and the effect of operational factors on the performance of each method by a numerical simulation model. There were three different strategies:
concentrated, which were soluble and insoluble CO$_2$ injection into the bottom aquifer, CO$_2$ injection into pay zone, and simultaneous CO$_2$ injection into aquifer and pay zone. The result was that simultaneous injection into aquifer and pay zone leads to higher final oil recovery in EOR schemes.

3.2. Modifications to shale CO$_2$ processes

Due to the low porosity of shale, capillary forces are not negligible. Furthermore, adsorption has to be carefully considered due to a large specific area in shale. Pu and Li [26] gave a new formulation that includes the capillary force and adsorption through pore size distribution. A local density optimization algorithm was used to the adsorption model. In the Bakken field, the results of their investigations reduced the soaking time of the CO$_2$ huff “n” puff process and increased the 18% OOIP ultimate recovery.

3.2.1. Shale heterogeneity needs to be considered

Most of the unconventional reservoirs are heterogeneous, which influences the application of the huff “n” puff method. Chen et al. [27] studied the relationship between the reservoir heterogeneity and CO$_2$ huff “n” puff recovery through running simulations in the Elm Coulee Field of the Bakken. Shale heterogeneity had a significant negative impact, reducing the final recovery rate of the well.

3.2.2. There have not been large-scale CO$_2$ sequestration projects with shale

Large-scale demonstrations to prove CO$_2$ storage capability and capacity for very long periods of time in shale have not yet occurred [28]. According to Global CCS Institute [29], only 15 large-scale projects on CO$_2$ storage are taking place around the world with CO$_2$ capture capacity volumes ranging from 0.7 to 7 million tons per annum (Mtpa) in countries such as Norway, Algeria, Canada, and the United States. These do not include smaller projects that use CO$_2$ injection and end up sequestering smaller volumes, i.e., CO$_2$ EOR projects.

3.2.3. Improving CO$_2$ sweep efficiency

To maximize the effectiveness of CO$_2$ sequestration and adsorption in shale, it is important for the injected carbon dioxide to come in contact with as much reservoir volume as possible, a phenomenon known as sweep efficiency. Again, not enough CO$_2$ sequestration projects in shale formations have taken place and been monitored to show what the most effective conditions are to keep carbon dioxide sequestered. An increase in recovery rate from CO$_2$ injection under specified conditions can be used to estimate the optimum requirements to achieve utmost levels of sweep efficiency, but this is not necessarily the ideal condition for sequestration.

The available knowledge suggests recovery factors increase drastically when carbon dioxide is injected around minimum miscible pressure (MMP) that is around 1500 psi [30]. MMP can change by a few percentages depending on reservoir pressure, permeability, heterogeneity, and pore geometry.
One of the advantages of carbon dioxide is that its MMP is much lower than other gases; therefore, CO$_2$ MMP injection is possible under a wide range of reservoir pressures [31]. At around MMP, carbon dioxide and oil are miscible which leads to a zero entry capillary pressure. This allows carbon dioxide to enter the oil filled tight pores of shale and increase sweep efficiency and storage with high displacement efficiency. Also, the required soaking time, the time needed for injected gas to pierce and spread throughout the formation, appears to have a significant effect on sweep efficiency because of exceedingly low permeability of shale formations. Longer shut-in periods after CO$_2$ injection show higher oil recoveries that indicate a greater sweep efficiency [28].

3.3. Carbonate potential

In the South Sumatera Basin, 98 carbonate oil fields represent 59% of total original oil in place (OOIP) [32]. A study ranked these reservoirs based on CO$_2$ EOR and sequestration.

3.3.1. Challenges in carbonates present opportunities in CCS

Carbonate hydrocarbon reservoirs remain poorly understood; opposed to other storage sites, carbonates are likely to be hydrophobic (2/3rd of the world’s carbonate reservoirs are oil wetting). CO$_2$ dissolution in the oil phase is orders of magnitude higher than its solubility in brine as seen in Figures 4 and 5. In the context of CO$_2$ sequestration in carbonate hydrophobic storage sites, dissolution of CO$_2$ in the oil phase is favorable for the long-term CO$_2$ storage in comparison with free supercritical CO$_2$ storage or CO$_2$ dissolution in brines.

**Figure 4.** Pressure dependence of CO$_2$ dissolution in an oleic phase in 71°C [33].
4. Economics

One of the primary challenges facing CCS and CO₂ EOR is the cost of trapping and delivering CO₂. Large-scale injection of CO₂ for any purpose, can only reach its full potential when a supply chain and infrastructure are established, and most locations do not have access to a preexisting CO₂ infrastructure [16, 35].

For instance, while the impermeable shale barriers in an Illinois Basin are a perfect seal for a long-term sequestration of CO₂, the absence of a CO₂ delivery infrastructure, despite local electrical power facilities emitting over 255 [20] metric tons of CO₂ annually, still overcomes all the scientific potential in the area. The same scientific potential could allow low-temperature oil reservoirs to become sequestration targets, and to increase the local CO₂ storage capacity 20 times, at the same time, to enhance the oil recovery by another 6–18% (360–1100 MMSTB) [21].

In one case in the Gazran field, the costs to acquire CO₂ were approximately 11$/metric ton, with recycling costs of approximately 8$/metric ton [16]. In other areas, such as West Texas, prices can be as high as $40/ton with 18 billion tons of CO₂ required, making it very difficult to initiate large-scale CO₂ projects without a proper supply chain. Ghomian et al. [36] estimated that the total costs of CO₂ sequestration are in the range of $40–$60 per ton of CO₂ stored, primarily due to the costs of CO₂ capture and compression. In cases where a proper CO₂ infrastructure can be created, CO₂ transported via a pipeline with rates above 10 million tons of CO₂ per year often cost less than $1/metric ton of CO₂ per 100 km, with lower flow rates costing as much as double that amount [34]. This suggests that once a basic infrastructure has been created, the capture cost of CO₂ will become the limiting factor in CCS and CO₂ EOR projects.
4.1. Coal bed

No matter how efficiently CO\textsubscript{2} ECBM and CO\textsubscript{2} sequestration works when CO\textsubscript{2} is readily available, economic problems cannot be ignored. Robertson [37] provided the economic analysis of CO\textsubscript{2} sequestration and CO\textsubscript{2} ECBM of the Powder River Basin in Wyoming. He evaluated three production scenarios (no gas injection, flue gas injection, CO\textsubscript{2} injection). Strategies were analyzed using a discount rate of 10% and the rate of return on investment. A Monte Carlo model was used to analyze the CO\textsubscript{2} injection method and the mean value of the CO\textsubscript{2} injection scenario (Figure 6). It was found that for the mean case, a cost of CO\textsubscript{2} of approximately $4.81/Mg (or $4.81/metric ton) is required to maintain the economic viability [35].

![Figure 6. Distribution and mean value of the cost of CO\textsubscript{2} separation/capture required to yield a 10% rate of return [35].](image)

Robertson also suggested that separating CO\textsubscript{2} from flue gas and injecting it into the unminable coal zones of the Powder River Basin seam, while currently uneconomical, can increase recovery of methane by 17% and could sequester over 86,000 tons CO\textsubscript{2}/ac [35].

A 2009 economic analysis by Gonzalez et al. investigated the effectiveness of CO\textsubscript{2} EOR and sequestration on coal beds of different initial permeability values and determined that CO\textsubscript{2} storage was often quite economical on wells of moderate permeability (10 milliDarcy) and high permeability (100 milliDarcy). In their study, none of the low permeability cases were economical. It is worth to mention that high-rank coals (those containing higher levels of carbon) showed the strongest economics [38].

5. Injection and sequestration

Unlike in the oil industry where the inability to recover injected resources is often a cause for concern and additional economic strain, CCS inherently requires the permanent sequestration of CO\textsubscript{2} in the given reservoir. These conflicting intentions will need to be overcome.
for economic purposes if CCS and CO₂ EOR are to become major players in the fight against climate change. Once these challenges have been overcome, the effectiveness with which CO₂ can be sequestered into different formations becomes a major point of importance.

Examples of the effectiveness of CO₂ sequestration are fairly common. Yamaguchi et al. [39] investigated the Ishikari Coalfield in Japan, where a multi-well test was able to inject 600 tons of CO₂ with an estimated 96% of the CO₂ being successfully adsorbed into the coal bed. Mavor et al. [40] analyzed a project by the Alberta Research Council which operated a two-well pilot test, where they determined that the increase in CO₂ injectivity (owing ballooning and water saturation reduction) was able to overcome injectivity losses due to swelling. Results were greatly improved by reducing injection periods, which allowed for adsorbed gas in the coal bed to finish swelling and for CO₂ to diffuse throughout the reservoir. These results were mirrored by Wan and Sheng [41], who determined that in fractured reservoirs, cyclic gas injection could increase oil recovery to 29%, while primary production only produced about 6.5% of OOIP [39]. Sheng and Chen [42] compared CO₂ and water flooding and were able to achieve superior results for CO₂ injection both in the case of flooding and huff “n” puff scenarios, with the best results (production of 32.46% OOIP) occurring using the huff “n” puff method.

5.1. CO₂ EOR in gas condensate wells

Higher densities of CO₂ relative to the native gas condensate cause CO₂ to migrate downward; with an increase of viscosity, CO₂ will displace the hydrocarbon gas phase. CO₂ EOR is very effective in light and medium gravity reservoir oils, in addition to being effective at recovery of gas condensates [43]. The dissolution of CO₂ into the oil decreases its interfacial tension; this creates a chance for the capillary force to enhance the recovery of the residual oil. This aspect heavily depends on the pressure and thus the depth. The properties of depleted gas/condensate reservoirs make them favorable for repressurization and enhanced gas recovery using CO₂ [41].

6. Possible geomechanical problems

EOR through CO₂ sequestration provides great opportunities for improving hydrocarbon recovery and the reduction of the greenhouse effect. Yet there are still problems about CCS that need to be addressed. A study on a pressure-depleted gas reservoir in the southern North Sea provided insight on CO₂ sequestration in depleted hydrocarbon reservoirs [60]. Their sequestration led to multiple geomechanical problems during drilling, completion, and CO₂ injection. These depleted reservoirs have a narrow window of drilling mud weights that will not result in reservoir problems, and well completions can be affected by potential solid flow back when the injection of CO₂ is interrupted, while the temperature changes near the wellbore can lead to thermal fracturing and reactivation of faults. CO₂ sequestration can sometimes require drilling additional injection wells, which can be a problem with a narrow mud weight window because of the increased chance of a wellbore collapse.
The narrow mud weight window can make it nearly impossible to avoid falling out of the ideal range of mud weights, leading to a number of risks and an increase of nonproductive time and additional costs. During the injection stage, if there are problems with CO₂ supply, resulting in an interruption of CO₂ injection, solids will flow back into the well, resulting in a risk of rock failure or erosion of a pipeline.

Well integrity is the achievement of fluid containment and pressure containment within the well throughout its whole life cycle. The CO₂ injection can lead to the corrosion and degradation of the injection tubing, injection casing, and cement and packer material. The trickiest part is keeping the well leak-free. A CO₂ sequestration well has to be designed for over 40 years of continued well integrity. Some potential methods of protecting well integrity include the injection of supercritical CO₂ fluid, as it is dry and noncorrosive, protecting a well for a much longer period [44]. Usage of supercritical CO₂ unfortunately, increases costs and can increase issues with temperature changes, which can hydraulically and thermally fracture a rock in a near-wellbore region. This risk can be mitigated by keeping the fluid pressure that acts on a caprock outside of its fracturing pressure. Most other well failure problems can be reduced by keeping a well straight instead of inclined [60].

6.1. Offshore leak issues

Offshore injections of CO₂ for EOR and sequestration lead to alterations and deformations of caprock, affecting seal integrity. A break in a cap rock can result in a large burst of CO₂ from a reservoir and ultimately the seabed. When evaluating long-term caprock integrity, it is important to note the intrinsic caprock properties, chemical conditions at reservoir/caprock interface, and injection-induced pressure perturbation [61].

The caprock properties to look for are fracture normal stiffness, bulk concentration, and carbonate-forming cations. The enhancement or degradation of a caprock is related to the reduction and widening of microfracture apertures. During an injection process, initial mineral trapping takes place, which can have a significant impact on maintaining initial CO₂ injectivity and can delineate and partially self-seal plume boundaries while also reducing caprock permeability. Many CO₂ migration and sequestration processes in saline aquifers are equally applicable to CO₂ flood EOR in shale-capped water-wet oil reservoirs [21]. The CO₂ storage capacity is inverse proportional on reservoir permeability, which, in pure sequestration scenarios with high injection pressure, benefits from an increased storage and delayed migration, providing a noncompromised caprock performance.

Injection could also lead to pressures exceeding the formations natural fracturing pressure, resulting in the reactivation of a fault or the reservoir rock becoming hydraulically and thermally fractured. This creates a potential breach in the caprock that prevents CO₂ migration to the surface or flow into an adjacent formation [60]. An injection is followed by a change of reservoir temperatures that result in expansion and contraction of materials and ultimately result in changes of the field stresses, which creates a risk of breaching the caprock over time.

All geomechanical problems impose a great risk on CO₂ storage, which means the caprock integrity must be addressed when selecting a storage well site. The sealing efficiency is dependent
on many factors, including caprock, well cement, capillary threshold pressure, and chemical reactivity to CO$_2$. A proper geological evaluation is required to investigate the possible paths for CO$_2$ migration to the surface through the faults and fractures. Well sites with microseismic activity are generally poor candidates for the long-term containment of CO$_2$. Topography has to be addressed in the same manner, in the case of CO$_2$ leakage; the surface has to be well ventilated to prevent an accumulation of CO$_2$ cloud.

6.2. Risks and examples of CO$_2$ leakage

Equipment degradation is a big problem in abandoned wells, as well as currently operating wells. Individual wells have to be monitored in order to spot a leakage of CO$_2$ through the annulus of a wellbore. Leakage can result in not only a migration of CO$_2$ to the surface but also a contamination of surrounding reservoirs and aquifers [43]. This can happen because of wellbore expansion and contraction due to temperature and pressure changes.

Nygaard et al. [45] wrote a paper regarding wellbore well leakage and found that 95 out of 1000 wells near Wabamun Lake in Alberta identified as potential leakage pathways caused by an immediate caprock penetration. This sort of issue is common in poorly plugged wells which can leak CO$_2$ at the cement-rock interface or through a cement plug. Any mechanical load during a completion or stimulation can affect the integrity of the cement, in addition to corrosion and chemical reactions near the wellbore. Issues such as those found in this study must be considered during the life and abandonment of well to ensure a reliable seal.

6.3. Actions to prevent future CO$_2$ leakage

There are multiple options for sealing abandoned wells, but all of them require at least 8 m of cement inside the casing. Most abandoned wells after 1995 have sufficient integrity. In order to improve the seal integrity, it is suggested to remove the casing steel from abandoned wells before the final cement plug, and an injection of the CO$_2$-resistant polymer is executed [44]. The cement samples from 30- to 50-year-old wells kept a good sealing integrity and prevented leakage of CO$_2$, even though they contained a degree of carbonation [10]. It is not suggested to squeeze the cement into an opening in the casing, but a melted alloy can fill most openings, and its expansion will mitigate microfissures [44].

CO$_2$ injection affects the mineralogy and structural heterogeneity of the reservoir, which will have an impact on the porosity, permeability, and storage stability of the well. Better predictions of reservoir response to CO$_2$ injection are a necessary step in the evaluation of possible long-term sequestration of CO$_2$. A well-proven method for CO$_2$ testing is Hassler cell core testing, but unfortunately, there is no standard protocol for CO$_2$ testing, which can lead to errors in results [10].

6.4. Selection of readings on stress and possible leakage in ECBM wells

As ECBM reservoirs often do not have a standard caprock to prevent leakage, their long-term viability as CO2 sequestration targets must be carefully considered. Numerous papers have explored this question and are briefly listed and summarized below:
• In the report of Myer [46], geomechanical factors that risk CO\(_2\) leakage in sequestration, as well as the risk of CO\(_2\) leakage from drilling and completion, production, and repressurization, are discussed.

• Mitra and Harpalani [47] investigated the matrix strain resulting from a CO\(_2\) injection.

• Chen et al. [48] investigated how the effective stress factor and methane CO\(_2\) counter-diffusion work on the CO\(_2\) recovery using a finite element model that coupled coal deformation, gas flow, and methane-CO\(_2\) counter-diffusion. Through their study, it was found that permeability loss/gain is influenced by effective stress and methane CO\(_2\) counter-diffusion and that the gas pressure distribution is related to gas composition [47].

• Fathi and Akkutlu [49] investigated counter-diffusion and competitive adsorption effects according to the new one-dimensional theoretical model they created. Compared with the conventional model, they created a new triple porosity dual permeability multi-continuum model to describe the gas release from macro-pores and micro-pores to the fracture.

6.5. CO\(_2\) monitoring

Due to the quantities of CO\(_2\) being sequestered in large CCS projects and the importance of keeping that CO\(_2\) permanently underground, monitoring is a very important part of any CCS project. 3D seismic survey has proven to be effective at monitoring CO\(_2\) storage but is prohibitively expensive. Gasperikova and Hoversten [50] investigated using a combination of gravity inversion, electromagnetic (EM), and amplitude vs. angle (AVA) monitoring analysis to detect changes in CO\(_2\) saturation. Gravity inversion detects density changes in the injected layer. EM and AVA can be used to estimate CO\(_2\) saturation changes. Macdonald [51] provided field and lab measurements of CO\(_2\) using Raman spectroscopy, which improved monitoring of the prevised amounts of CO\(_2\) dissolved in reservoir brine. Through the Saptharishi and Makwana [52] study, various monitoring techniques are summarized, which include but are not limited to techniques for coal beds.

6.6. Risks of CO\(_2\) injection, possible failure modes

Carbon dioxide storage is not a risk-free task. As years go by after CO\(_2\) has been injected into a formation, it is possible for the CO\(_2\) to begin migrating upward and leak out of the ground back into the atmosphere through openings in the caprock or fractures, faults, and poorly completed preexisting wells [53]. This problem can be prevented or reduced if the formation of interest for CO\(_2\) storage has a caprock with ideal qualities.

An ideal caprock is a layer of the formation with very low permeability that can prevent oil and gas from migrating upward and out of the reservoir formation. In any case of CO\(_2\) storage, a thick shale layer is the most desirable type of caprock. Due to shale’s extremely low permeability, causing a more tortuous flow path, CO\(_2\) migration vertically is tremendously limited [54]. The degradation of cement and metal casing with a presence of CO\(_2\) is currently a topic that needs extensive investigation [52]. As the Figure 7 shows, there are five possible leakage pathways in an abandoned well. Label a and label b are the pathway between casing
and cement wall and plug, respectively. Label c shows leakage through cement plugs. Label d represents leakage through casing. Label e shows leakage through the cement wall. Label f represents leakage pathways between the cement wall and the formation [52].

The Sleipner Project, located in Norway, is currently storing more than 2700 tons of CO$_2$ per day below an extensive and thick shale layer [27]. Monitoring the injected CO$_2$ during the past 13 years is showing that the gas has spread out nearly two square miles below the shale layer without moving upward or leaking out of the reservoir storage [55]. This is one of the most significant evidence that proves how effective shale formations can be as CO$_2$ storage reservoir and caprock, where carbon dioxide will be trapped and immobile. In short, the ultimate geological storage reservoir should have sufficient capacity, be a thick shale layer acting as a caprock simultaneously, and be a stable storage environment maintaining the original characteristics of the reservoir.

6.6.1. Overcoming the high risk of CO$_2$ leakage in carbonate reservoirs

Carbonate reservoirs do not generally possess an impermeable boundary or caprock, and therefore permanent trapping of CO$_2$ through geomechanical means is unrealistic [57]. Solubility storage decreases potential leakage in carbonate formations, as the dissolution of CO$_2$ into water promotes mineralization, but this will need to be studied further before carbonate reservoirs can be relied upon to properly sequester large volumes of CO$_2$ [56].

Agada et al. [57] did extensive research on how fracture network geometry affected oil recovery and CO$_2$ storage in carbonate reservoirs. They noted that many of the problems associated with high fracture-matrix connectivity, such as bypassing of oil, early water breakthrough, and rapid CO$_2$ migration, could be mitigated by foam flooding. Sehbi et al. [58] proposed a low injection rate, longer in-reservoir CO$_2$ retention time, and good pore structure to improve the efficiency in carbonate reservoirs. Carbonate formations showed an increase in effective permeability resulting from chemical dissolution in the matrix, thus enhancing pore connectivity [59].

Figure 7. CO$_2$ leakage pathways.
6.6.2. Examples of natural carbonate sequestration

The Colorado Plateau-Southern Rocky Mountains region contains natural CO\textsubscript{2}, which has been discovered during exploration for oil and gas fields (Figure 8), thus providing a natural laboratory for studying the effects of long-term, subsurface CO\textsubscript{2} storage in carbonate reservoirs. These laboratories yield information such as that injecting CO\textsubscript{2} separated from flue gases ensures the subsurface migration path is long, thereby yielding optimal sequestration. Despite the number of carbonate CO\textsubscript{2} reservoirs in the region and active flux of CO\textsubscript{2} to the surface, no hazards from CO\textsubscript{2} surface accumulations are known. The nature and rate of CO\textsubscript{2} surface leakage in carbonate formations are still unknown [60].

![Figure 8. Synthesis of data relating to CO\textsubscript{2} fluxes and concentrations around the Colorado Plateau [8].](image)

6.7. Additional potential concerns

Despite the many benefits of CO\textsubscript{2} EOR and CCS programs, it must be remembered that these are complicated projects being undertaken in complex geological environments. A 2004–2008
project in Algeria stored over 2.5 [61] million tons of CO\textsubscript{2} in a carboniferous sandstone reservoir. During the shut-in process, the CO\textsubscript{2} injection was unexpectedly interrupted, and the wellbore pressure went lower than the reservoir pressure, risking rock failure, sand production, and possible blowout.

Potential concerns also include preventing potentially catastrophic failure of the reservoir seal. For instance, if injection pressures exceed the breakthrough pressure of the sealing caprock, the CO\textsubscript{2} would break through and risk flowing back to the surface [60]. Reactive transport modeling shows that for a typical shale caprock, geochemical processes continuously improve isolation performance, and geomechanical processes first rapidly degrade and then improve isolation performance over time. There is a possibility for a counterbalancing of geomechanical and geochemical effects, but they must be carefully monitored [62].

Some issues are more minor, not directly threatening the safety of the operation, but nonetheless affecting the economics of a combined CO\textsubscript{2} EOR and CCS project. Bou-Mikael [14] wrote about the performance of a CO\textsubscript{2} flood at Port Neches in the Gulf of Mexico, with a partnership of Department of Energy and Texaco E&P. The CO\textsubscript{2} flood underperformed [13], with 500 bbl/day instead of 800 bbl/day; with this underperformance was attributed to the following reasons: reservoir characterization, oil saturation, water blockage, and wellbore mechanical problems. After a careful evaluation of the project, it was determined that in ideal circumstances and if related criteria are met, CO\textsubscript{2} injection can accelerate production two to three times compared to unassisted primary production [13].

6.7.1. Potential coalbed problems

Coal beds, despite offering unique opportunities, also offer unique challenges. In particular the coal matrix swells during CO\textsubscript{2} adsorption. Coal matrix swelling can cause reductions in permeability. Bustin et al. [63] experimented on the volumetric strain from three western Canadian coals and found that a mixture of N\textsubscript{2} and CO\textsubscript{2} injection would improve CO\textsubscript{2} injection rates greatly but that CO\textsubscript{2} sequestration capacity decreased wildly. However, pure CO\textsubscript{2} injection could cause the reduction of permeability by two orders of magnitude. The applicability of the CO\textsubscript{2}-ECBM process in any coal seam is mainly governed by the seam’s permeability and its adsorption process [62]; therefore, these concerns must be explored further.

7. Conclusion

Numerous studies support the potential of major sequestration projects, and due to the negative impacts of atmospheric CO\textsubscript{2}, CCS will continue to be an important part of protecting our environment. While EOR through CO\textsubscript{2} sequestration has proved to be valuable, there are still challenges that need to be addressed in the future. Reservoir properties must continue to be carefully considered for all CCS projects due to their impact on successful EOR and CO\textsubscript{2} sequestration.

The major challenges currently facing CCS projects are primarily those of economics and transportation. Limited CO\textsubscript{2} transportation supply chains act as a barrier for CO\textsubscript{2} EOR utilization
in the oil industry. When this barrier has been removed and a large network of CO\textsubscript{2} capturing mechanisms have been created, it will open the petroleum industry to a breadth of new possibilities both in terms of improved recovery and environmental sustainability.

For the purposes of having a significant impact on atmospheric CO\textsubscript{2} levels, the simple merging of CO\textsubscript{2} EOR and CCS may not be enough. In every reservoir type, in every circumstance, there are diminishing returns as far as incremental production as additional CO\textsubscript{2} is injected into a well. As such, as long as CO\textsubscript{2} remains an expense, rather than a revenue stream, the full potential of CCS will not be realized. In the meantime, however, there are numerous projects that hold a good deal of promise and are economical under current conditions due to the benefits of CO\textsubscript{2} injection on ultimate hydrocarbon recovery.

7.1. Closing notes on shale reservoirs
Shale reservoirs still hold a great deal of promise for CCS and CO\textsubscript{2} EOR, as the benefits for production are significant, and the formations themselves provide excellent seals against any risk of CO\textsubscript{2} migration. Unfortunately, a great deal of research and monitoring is still required in order to ensure that shale beds maintain the quality of their seals over time and to maximize CO\textsubscript{2} sequestration. Knowledge gaps such as lack of information on available storage capacity, lack of formation and reservoir data that specifies favorable sequestration settings, understanding long-term CO\textsubscript{2} interaction in shale, and testing different strategies for CO\textsubscript{2} injection and well patterns to achieve efficient carbon dioxide sequestration and EOR still exist [52]. Many questions regarding this topic will remain unanswered until additional, large, in situ field tests take places.

7.2. Closing notes on carbonate reservoirs
The future of CO\textsubscript{2} EOR and sequestration in carbonate reservoirs will steadily improve due to the statistical data being acquired from existing field tests. The United States’ carbonate formations provide the foundation for CO\textsubscript{2} injection in carbonate reservoirs [64]. The Bati Raman reservoir provides a significant opportunity to further carbonate CO\textsubscript{2} EOR operations. Sahin et al. [65] state this reservoir could easily yield a billion dollars in revenue as a CO\textsubscript{2} EOR project. Hydrocarbon fuels can supply relatively pure CO\textsubscript{2} for EOR allowing the byproducts of the industry’s previous production to add in new production while also creating a more environmentally friendly outcome. CO\textsubscript{2} that cannot be used for EOR can be stored in depleted carbonate formations, thus furthering the climate-friendly initiative [66]. Recent estimates of future CO\textsubscript{2} demand suggest that large volumes will be required to meet the promise of next-generation EOR including the development of residual oil zones [7].

7.2.1. Specific challenges in carbonate reservoirs
As previously discussed, dissolved CO\textsubscript{2} injection is recommended for reactive fractured formations and formations with uncertain caprock integrity [7]. The challenges of the carbonate pinnacle reef data analysis are as follows: an increase in pressure with CO\textsubscript{2} injection, the
presence of multiple reservoir fluids, and unique CO$_2$ phase behavior due to changing pressure and temperature.

Izgec et al. [1] discuss challenges of mineral trapping, the effects of changing rock properties, and the residual impact on CO$_2$ CCS in carbonate reservoirs. Puon et al. [67] state that other challenges in carbonate formations include CO$_2$ tendency to bypass a large percentage of pore volume, yielding an early breakthrough, and reductions in recovery efficiency. As a result, CO$_2$ flooding is not economically feasible without improved mobility control. Several mobility control methods have been attempted with limited success; therefore, concepts for CO$_2$ mobility control are required to increase the overall recovery efficiency and economics in carbonate reservoirs [66]. Eke et al. [8] suggested the injection of denser CO$_2$ saturated brine in carbonate formations, which should be capable of eliminating much of buoyancy force. Thus, CO$_2$ brine surface mixing strategy is recommended due to the enhancement and secure storage of CO$_2$ in subsurface carbonate formations.

8. Suggestions for future study

For sandstone and coalbed reservoirs, the last major remaining barrier to large-scale implementation of CO$_2$ EOR and CCS is the economic burden of CO$_2$ capture and transportation. Research into improving capture and transport techniques, as well as how to structure intelligent government incentives, will go a long way in increasing CO$_2$ sequestration rates.

Unlike with sandstone and coalbed reservoirs, the primary barrier to CO$_2$ sequestration in shale reservoirs is a lack of research and monitoring work after CO$_2$ injection. The lack of research is in fact only aggravated by the lack of monitoring and in situ data.

At last, the study of CO$_2$ sequestration in carbonate reservoirs needs to expand to include the effects of CO$_2$ on carbonate rock properties. Issues ranging from an early breakthrough, to low sweep efficiency, to structural problems within the formation, all, limit the viability of large-scale carbonate projects. Future experiments will need to be performed using a high-pressure carbonate core flooding system optimized for use within different lab apparatus so that experiments can be conducted to better understand the complications and benefits of supercritical CO$_2$ [57].

There is no doubt that CCS and CO$_2$ EOR/ECBM will play a major role in the future of the energy industry. However, besides the economic issues with many CO$_2$ implementations, legal risks must also be considered as well. Unitization is important for CO$_2$ EOR in order to avoid the trespass claim [68]. In addition, different states and regulators treat CO$_2$ differently, sometimes as a pollutant, other times as a natural gas [67]. Despite not being research-based concerns, the legal climate of the United States must also change for a truly successful CCS program to take hold and to make the best use of what could become a CO$_2$ revolution.
Nomenclature

AVA Amplitude vs. angle
bbl barrel
CCS Carbon capture and storage
CO₂ Carbon dioxide
DOE Department of Energy
ECBM Enhanced coalbed methane
EM Electromagnetic
EOR Enhanced oil recovery
MMP Minimum miscible pressure
MPZ Major pay zone
ROZ Residual oil zone
SACROC Scurry Area Canyon Reef Operators
SECARB Southeast Regional Carbon Sequestration Partnership
SWP Southwest Regional Partnership for Carbon Sequestration
TZ Transition zones

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References


