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Chapter 9

Microscopic Investigations on Woody Biomass as Treated with Ionic Liquids

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Abstract

Woody biomass is one of the most promising renewable alternatives to fossil resources. However, some physical and chemical treatment is required to convert their chemical components into biofuels and valuable chemicals because of their low degradative properties. Recently, there has been considerable interest in ionic liquid treatment for biorefinery, and many fundamental studies on the reactivity of wood with ionic liquids have been performed from a chemical and morphological point of view. This chapter highlights the findings regarding morphological and topochemical features of wood cell walls in the degradation process as a result of ionic liquid treatment. Bright-field microscopy and scanning electron microscopy have revealed the swelling behavior of cell walls and the detailed ultrastructural features of wood tissues treated with ionic liquid. Polarized light microscopy and confocal Raman microscopy have clarified the changes in cellulose crystallinity and distribution of chemical compositions such as polysaccharides and lignin during ionic liquid treatment at the cellular level.

Keywords: Biorefinery, Cell wall, Liquefaction, Ionic liquid, Wood

1. Introduction

The efficient use of lignocellulosics has been an important approach to prevent exhaustion of fossil resources and global warming caused by increasing emissions of greenhouse gases. Among the various types of lignocellulosics, woody biomass is regarded as a promising resource because it is carbon neutral, abundantly available in many regions, and does not compete with agricultural production. Wood cell walls decompose to form persistent organic complexes mainly composed of cellulose (40–50%), hemicellulose (25–35%), and lignin (18–35%) [1, 2]. To convert
their chemical components into transportable biofuels and chemical feedstocks, it is necessary to develop the processing technology to break their rigid structure. To date, various conversion methods such as acid hydrolysis [3–5], enzymatic saccharification [6, 7], pyrolysis [8–10], and supercritical or sub-critical fluid treatment [11–13] have been investigated. However, practical methods have not yet been established.

Ionic liquids are defined as organic salts with low melting points, and have many advantages including negligible vapor pressures, chemical and thermal stability, non-flammability, low viscosity, and reusability [14–16]. In addition, they can dissolve a wide range of organic and inorganic substances [14]. Ionic liquid treatment is attractive as a new conversion technology for woody biomass. **Figure 1** shows the typical cations and anions found in ionic liquids. There are an infinite number of combinations of cations and anions, and their physical properties, such as melting point, viscosity and dissolving power, can be easily changed by altering the combination. This is why ionic liquids are called “designer solvents” [17].

![Figure 1](image)

**Figure 1.** The structures of typical cations and anions in ionic liquids.

Recent studies revealed that certain types of ionic liquids can liquefy cellulose [18–20], lignin [21], and even wood cell walls [22–26]. Using ionic liquids as the solvent to process woody biomass, many fundamental studies on the reaction behavior of wood in ionic liquids have been carried out focusing on the chemical processes [27–31]. However, woody biomass is a very inhomogeneous composite at the cell level. Wood comprises various types of tissues such as the tracheid, wood fibers, vessels, and parenchyma. In addition, wood cell walls consist of several layers: a compound middle lamella (middle lamella + primary wall; CML) and a secondary wall (S), which is generally composed of $S_1$, $S_2$, and $S_3$ sublayers. The chemical components and distribution vary depending on the wood species, types of tissues and their layers [32]. Therefore, to improve the chemical conversion process using ionic liquids, a better understanding of the effects of ionic liquid treatment of wood, such as the interaction of wood with ionic liquids at the cell level and the deconstruction behavior of various types of tissues in ionic liquids, are required.

In this chapter, we focus on morphological and topochemical studies on the liquefaction of wood in ionic liquids, especially 1-ethyl-3-methylimidazolium chloride ([C2mim][Cl]) and 1-ethylpyridinium bromide ([EtPy][Br]), using various microscopy techniques. [C2mim][Cl] and [EtPy][Br] are known as the ionic liquids which can preferentially liquefy cellulose [25] and...
lignin [31], respectively. Bright-field microscopy and polarized light microscopy were employed to determine the swelling and decomposition behaviors of wood cell walls and the state of cellulose crystallinity during ionic liquids treatment. Scanning electron microscopy (SEM) was used to observe the detailed ultrastructural changes in various wood tissues treated with ionic liquids. Confocal Raman microscopy was employed to examine the changes in chemical components including polysaccharides and lignin at the cellular level and to visualize their distribution on the cell walls during ionic liquids treatment.

2. Application of microscopy techniques to examine wood liquefaction

2.1. Light microscopy analysis

Morphological features of wood cell walls during the liquefaction process in ionic liquid were determined by light microscopy [33–38]. Figure 2 shows bright-field microscopy and polarized light microscopy images of Cryptomeria japonica, which is the most common softwood species in Japan, treated with [C2mim][Cl] at 120°C for 0 h and 24 h. All the cell walls swelled after 24 h of [C2mim][Cl] treatment (Figure 2c). Although the cell walls of tracheids in latewood (summerwood; formed late in the growing season) were disordered and distorted after the treatment, those in earlywood (springwood; formed early in the growing season) were barely changed. The polarized light microscopy studies show that the brightness from the birefringence of cellulose in both earlywood and latewood was decreased after treatment (Figure 2d). These results imply that the crystalline structures of cellulose in wood are amorphized before the wood cell walls liquefy completely during [C2mim][Cl] treatment.

![Figure 2](image-url)
To study the detailed swelling behavior of tracheids arising from [C2mim][Cl] treatment, we performed time sequential measurements of the cell wall area, cell lumen area, and the total of cell lumen and cell wall areas, in earlywood and latewood, in transverse sections (Figure 3). The cell wall area in earlywood increased only slightly at an early stage of [C2mim][Cl] treatment, whereas that in latewood increased significantly. After the initial swelling, the cell wall area in earlywood showed no further changes, whereas that in latewood increased gradually with prolonged treatment time. After 72 h of treatment, the cell wall area in earlywood and latewood had increased by 1.5 and 4 times, respectively. These results indicate that the swelling behavior of the tracheids of *Cryptomeria japonica* is different for the two morphological regions. In particular, tracheids in latewood are more prone to swelling and being broken after [C2mim][Cl] treatment.

Figure 3. Changes in cell wall area, cell lumen area, and total of cell lumen + cell wall area in earlywood and latewood of *Cryptomeria japonica* during [C2mim][Cl] treatment at 120°C [34].

Changes in the cell wall area of fibrous cells of various Japanese hardwood species during [C2mim][Cl] treatment were also measured (Figure 4). At the initial stages of [C2mim][Cl] treatment, the cell wall areas of all species increased rapidly. Thereafter, the cell wall areas of *Fagus crenata* and *Quercus mongolica* in both earlywood and latewood increased continuously with progressing treatment time. After 72 h of treatment, the cell wall areas of *Fagus crenata* in both earlywood and latewood had increased by 4 times, and those of *Quercus mongolica* in earlywood and latewood had increased by 3.5 and 5 times, respectively. However, the swelling behavior of two other species that were tested was different from *Fagus crenata* and *Quercus mongolica*. The cell wall area of *Quercus glauca* in latewood increased gradually with progressing treatment time, but not in the earlywood. The change in cell wall area in *Trochodendron aralioides* levelled off in both earlywood and latewood. After 72 h of treatment, the cell wall areas of *Quercus glauca* in earlywood and latewood had increased by 2 and 3 times, respectively, and those of *Trochodendron aralioides* in earlywood and latewood had increased by 1.5 and 2 times, respectively. Therefore, swelling behavior, including the swelling ratio and time-
dependent change of fibrous cells during [C2mim][Cl] treatment, differed according to the wood species and morphological regions, including earlywood and latewood.

Figure 4. Changes in the cell wall area of fibrous cells in the earlywood and latewood of various wood species during [C2mim][Cl] treatment at 120°C [35, 36].

**Figure 5** shows bright-field microscopy and polarized light microscopy images of *Cryptomeria japonica* before and after treatment for 72 h with [EtPy][Br] at 120°C. The cell walls of both earlywood and latewood were swollen by [EtPy][Br] treatment (**Figure 5c**). Although the cell walls in earlywood were well ordered during [EtPy][Br] treatment, those in latewood were partially dissociated. After 72 h of treatment, tracheids in earlywood and latewood were

Figure 5. Bright-field microscopy (a, c) and polarized light microscopy images (b, d) of transverse sections of *Cryptomeria japonica* before (a, b) and after treatment with [EtPy][Br] at 120°C for 72 h (c, d) [37].
swollen by 1.3 and 2 times, respectively. Therefore, the swelling efficiency of [EtPy][Br] is lower than that of [C2mim][Cl]; and morphological changes in wood cell walls differ for the different types of ionic liquids. In the polarized light micrographs, the brightness from the birefringence of cellulose showed no changes during [EtPy][Br] treatment (Figure 5d). This result implies that [EtPy][Br] treatment has no marked effect on the crystalline structure of cellulose in the wood cell walls.

2.2. Scanning electron microscopy observations

Ultrastructural changes in wood cell walls due to ionic liquid treatment were observed by SEM [34–38]. Figure 6 shows SEM images of various tissues of Cryptomeria japonica treated with [C2mim][Cl] and [EtPy][Br]. The dissociations and distortions of cell walls were observed in latewood after 72 h of treatment with both [C2mim][Cl] and [EtPy][Br] (Figure 6e, i). These changes were caused by significant swelling of tracheids in latewood. The magnified view of earlywood (Figure 6b, f, j) and latewood (Figure 6c, g, k), shows the CML (indicated by small arrows) disappeared in both earlywood and latewood after [EtPy][Br] treatment (Figure 6j, k) but was preserved after [C2mim][Cl] treatment (Figure 6f, g). These changes may be related to lignin distribution in wood cell walls. CML contains more than 50% lignin, whereas S$_2$ (indicated by arrowheads) contains about 20% lignin [39]. Our recent research revealed that lignin can be preferentially liquefied by [EtPy][Br] [31], which explains why the CML was liquefied more rapidly than S$_2$ by [EtPy][Br]. Figure 6d, h, and l shows the changes in bordered pits in earlywood. The torus (indicated by large arrows) in the pits was broken after [C2mim][Cl] treatment, while the warts on the surface of tracheids disappeared after [EtPy][Br].

![Figure 6](image-url)
treatment. Harada and Côté have reported that the torus is mainly composed of cellulose microfibrils [40]. In addition, Jansen et al. stated that warts are mainly composed of lignin and hemicellulose [41]. The differences in the reactivity of [C2mim][Cl] and [EtPy][Br] with these wood tissues derive from differences in their chemical components.

Figure 7 shows SEM images of various tissues of *Fagus crenata* treated with [C2mim][Cl] and [EtPy][Br]. Cracks occurred between the ray parenchyma (indicated by large arrowheads) and peripheral tissues after treatment with both ionic liquids. These changes are presumably due to the bonds between the ray parenchyma and tissues adjacent to it weakening after treatment as well as the difference in the swelling behavior. Figure 7b, f, and j show magnified views of the morphological features of wood fibers treated with [C2mim][Cl] and [EtPy][Br] which are similar to tracheids in latewood of *Cryptomeria japonica* (Figure 6g, k). Various changes were observed at vessel pits. Intervascular pits (vessel–vessel pits) were occluded after [C2mim][Cl] treatment (Figure 7g), while ray-vessel pits were not occluded. Instead, their pit membranes (indicated by large arrows) were broken after the treatment (Figure 7h). However, the intervascular pits and ray-vessel pits showed no significant changes after [EtPy][Br] treatment (Figure 7k, l). These results indicate that morphological changes in the pits differed depending on the types of pits and ionic liquids used.

Figure 8 shows SEM images of transverse sections of two hardwood species treated with [C2mim][Cl]. Although wood fibers (indicated by arrowheads) of both *Fagus crenata* and *Quercus mongolica* showed significant deformation, the axial parenchyma cells (indicated by arrows) maintained their shapes (Figure 8c, d). These differences in morphological changes between the wood fibers and axial parenchyma cells are mainly attributed to their chemical
composition. Fujii et al. reported that the lignin content of the axial parenchyma cells is higher than that of the wood fibers [42]. In our previous papers, we showed that lignin is much more difficult to react with [C2mim][Cl] than cellulose and hemicellulose [25, 28]. The reactivity of [C2mim][Cl] with the axial parenchyma cells is lower than that with the wood fibers, thus the axial parenchyma cells did not collapse after [C2mim][Cl] treatment.

Figure 8. SEM images of transverse sections of Fagus crenata (a, c) and Quercus mongolica (b, d) before and after treatment with [C2mim][Cl] at 120°C for 72 h [35, 36]. Wood fibers and axial parenchyma cells are indicated by arrow heads and arrows, respectively.

2.3. Confocal Raman microscopy analysis

Raman spectra can reveal much information about functional groups, hydrogen and chemical bonds, and the surrounding environment. Raman spectroscopy is used to identify the chemical structure of a substance. Confocal Raman microscopy couples Raman spectroscopy with a confocal microscope to perform detailed analysis quickly. In recent years, confocal Raman microscopy has received attention as a new method of spectroscopic analysis for plant cell walls because of its characteristic advantages. It is non-destructive, has a high spatial resolution (approximately 0.3–2 μm), is not hindered by the presence of water [43, 44], and little-to-no sample pre-treatment is required. Several research groups have studied the chemical composition of native wood cell walls using this method [45–51]. In addition, it has been reported that confocal Raman microscopy is an effective tool to investigate topochemical changes in wood after pre-treatment for biorefinery [52–54].

We applied confocal Raman microscopy to determine the changes in chemical components and their distribution in wood at the cellular level during [C2mim][Cl] and [EtPy][Br] treatment [37, 38, 55, 56]. Figure 9 shows Raman spectra obtained from S2 of the wood fibers of Fagus crenata treated with [C2mim][Cl] and [EtPy][Br]. The characteristic Raman bands of lignin were observed at 370, 1331, 1459, 1600, 1657, 2848, and 2943 cm⁻¹, whereas those of the polysaccharides (cellulose and hemicellulose) were observed at 380, 436, 520, 1093, 1118, 1152,
1378, and 2895 cm$^{-1}$. These band assignments were based on previous studies [57–61]. Although all the band intensities of the polysaccharides sharply decreased during [C2mim][Cl] treatment, most band intensities of lignin barely changed, except for some bands (Figure 9a). For instance, the band intensity at 1657 cm$^{-1}$ was assigned to the ethylenic C=C bond in coniferyl/sinapyl alcohol units and γ—C=O in coniferyl/sinapyl aldehyde units decreased gradually. These results indicate that [C2mim][Cl] prefers to react with polysaccharides in S$_2$ of wood fibers rather than with lignin, despite the molecular structures of lignin changing partially during [C2mim][Cl] treatment. Regarding time-dependent spectral changes during [EtPy][Br] treatment, most band intensities of polysaccharides showed no significant changes, while those of lignin decreased with prolonged treatment time (Figure 9b). These spectral changes indicate that lignin is liquefied more readily than polysaccharides in [EtPy][Br].

Figure 9. Raman spectra for the S$_2$ of wood fibers of Fagus crenata after treatment with [C2mim][Cl] (a) and [EtPy][Br] (b) at 120°C for 0 h, 24 h, and 72 h. Lignin and polysaccharides are indicated by solid lines and dashed lines respectively [38, 56].

To study the changes in the distribution of chemical components in the wood cell walls over a wide range during ionic liquids treatment, Raman mapping analysis was applied on transverse sections. Raman mapping was performed on tracheids and wood fibers because these tissues are the main elements of Cryptomeria japonica and Fagus crenata, respectively. In the Raman images, bright areas indicate high concentrations of a specific chemical composition, whereas dark areas indicate low concentrations.
Figure 10 shows the results of time-sequential Raman mapping analysis of the distribution of lignin and polysaccharides of tracheids of Cryptomeria japonica and wood fibers of Fagus crenata during [C2mim][Cl] treatment. Within both wood species, the lignin concentration was higher in CML than in S2 while the polysaccharide concentrations were lower in CML than in S2. After treatment, changes in the distribution of chemical compositions were similar for the tracheids of Cryptomeria japonica and wood fibers of Fagus crenata. Although the lignin concentration in S2 decreased with increasing [C2mim][Cl] treatment time, the concentration in CML showed no significant change. These results indicate that lignin in CML has a high chemical resistance to [C2mim][Cl]. The differences in the reactivity of [C2mim][Cl] with lignin in each morphological region are attributed to many factors such as molecular structure, concentration, and penetrability of [C2mim][Cl]. Polysaccharide concentrations decreased significantly after 24 h of treatment. After that, most of the Raman signal derived from polysaccharides disappeared. Therefore, in fibrous cells of wood, polysaccharides were more rapidly liquefied by [C2mim][Cl] than lignin.

![Figure 10](image_url)

Figure 10. Raman mapping of transverse sections of tracheids of Cryptomeria japonica (A–I) and wood fibers of Fagus crenata (a–i) before and after treatment with [C2mim][Cl] at 120°C for 0 h, 24 h, and 72 h [55, 56]. The left panel shows bright field images of the measured position, the middle panel shows the distribution of lignin, the right panel shows the distribution of polysaccharides. Bright regions indicate high concentrations of specific chemical compositions, dark regions indicate low concentrations.
Raman mapping analysis was also performed on the wood samples treated with [EtPy][Br] (Figure 11). Although the lignin concentration in S$_2$ and in CML decreased with prolonged treatment time, the lignin in CML was preserved at relatively high concentration after 72 h of treatment in both tracheids of Cryptomeria japonica and wood fibers of Fagus crenata. Polysaccharide concentrations in both species decreased after 24 h of treatment. However, after 72 h of treatment, the polysaccharide concentrations in the tracheids of Cryptomeria japonica showed no significant changes, while those in the wood fibers of Fagus crenata continued to decrease gradually. These results imply that the reactivity of [EtPy][Br] with lignin is higher than with the polysaccharides in S$_2$ of fibrous cells. In addition, the liquefaction behavior of polysaccharides in fibrous cells during [EtPy][Br] treatment is different for different wood species.

Figure 11. Raman mapping on transverse sections of tracheids of Cryptomeria japonica (A–I) and wood fibers of Fagus crenata (a–i) before and after treatment with [EtPy][Br] at 120°C for 0 h, 24 h, and 72 h [37, 38]. The left panel shows bright field images of the measured position, the middle panel shows the distribution of lignin, the right panel shows the distribution of polysaccharides. Bright regions indicate high concentrations of specific chemical compositions, dark regions indicate low concentrations.

To gain insight into the reactivity of [C2mim][Cl] with various morphological regions, Raman spectra were acquired for the S$_2$ of wood fibers, the cell corner of wood fibers, vessels, and axial parenchyma cells of Fagus crenata (Figure 12). During [C2mim][Cl] treatment, all the polysac-
charide band intensities of $S_2$ of wood fibers and vessels decreased markedly, whereas those of the axial parenchyma cells decreased slightly. In addition, the intensity of the lignin band at 1657 cm$^{-1}$ decreased significantly for all the measured regions except for the sample of axial parenchyma cells. Overall, the Raman spectra for axial parenchyma cells were not changed by [C2mim][Cl] treatment compared with those for the other morphological regions. This result indicates that both the polysaccharides and lignin of axial parenchyma cells are difficult to react with [C2mim][Cl]. The tendency of the spectral changes agrees with the morphological changes observed by SEM (Figure 8).

Figure 12. Raman spectra focusing in the spectral region of 950–1750 cm$^{-1}$ for $S_2$ of wood fibers (a), cell corner of wood fibers (b), vessels (c), and axial parenchyma cells (d) of Fagus crenata before and after treatment with [C2mim][Cl] at 120°C for 0 h, 24 h, and 72 h [56]. Lignin and polysaccharides are indicated by solid lines and dashed lines respectively. The spectra were normalized based on the band of an aromatic ring vibration around 1600 cm$^{-1}$ (thick lines), which has superior chemical stability under [C2mim][Cl] treatment conditions, as an internal reference to visualize the spectral differences after treatment with [C2mim][Cl] clearly.

3. Conclusions

Using various microscopy techniques, the morphological and topochemical features of wood cell walls treated with ionic liquids were studied. During the processing of wood liquefaction in ionic liquids, the ultrastructure and chemical compositions of wood showed inhomogeneous changes at the cellular level. The interaction of ionic liquid with wood cell walls was quite different depending on the types of ionic liquids, wood species, tissues, and cell wall layers. These findings will serve to cultivate a better understanding of the liquefaction mechanism of woody biomass in ionic liquids and accelerate development of ionic liquid treatment for wood-based biorefinery.

For research in wood chemistry and anatomy, many microscopy techniques have been applied to investigate the characteristics of the cell walls. However, it is hardly possible to examine the chemical compositions and their distribution with nanoscale spatial resolution while at the
same time observing the ultrastructure such as ultrathin layers. The development of sensitive analytical methods in the wood cell walls for chemical information with much higher spatial resolution will open a new field of wood science and technology.

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