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# Mesoscopic Simulation Methods for Studying Flow and Transport in Electric Fields in Micro- and Nanochannels

Jens Smiatek<sup>1</sup> and Friederike Schmid<sup>2</sup>

<sup>1</sup>*Institut für Physikalische Chemie, Westfälische Wilhelms-Universität Münster, Münster*

<sup>2</sup>*Institut für Physik, Johannes-Gutenberg-Universität Mainz, Mainz  
Germany*

## 1. Introduction

In the past decades, several mesoscale simulation techniques have emerged as tools to study hydrodynamic flow phenomena on scales in the range of nano- to micrometers. Examples are Dissipative Particle Dynamics (DPD), Multiparticle Collision Dynamics (MPCD), or Lattice Boltzmann (LB) methods. These methods allow one to access time and length scales which are not yet within reach of atomistic Molecular Dynamics (MD) simulations, often at relatively moderate computational expense. They can be coupled with particle-based (*e.g.*, molecular dynamics) simulation methods for thermally fluctuating nanoscale objects, such as colloids or large molecules. This makes them particularly attractive for the application in microfluidic or nanofluidic research.

Mesoscale coarse-grained simulation models are characterized by (i) simple potentials and (ii) simple, often stochastic dynamics. On large length scales, chemical details like the explicit structure of the water molecules can be neglected, such that the water can either be replaced by a simpler model fluid ("explicit solvent") or altogether omitted ("implicit solvent"). A suspended particle undergoes stochastic collisions with the solvent resulting in Brownian motion. This process can be described by a Markovian process on the timescale of microseconds. Mesoscale coarse-grained techniques exploit this behavior. As long as the relevant conservation laws are obeyed (mass, momentum, energy, charge etc.), the actual solvent dynamics may be replaced by a computationally cheaper artificial dynamics. For example, transport mechanisms as well as structure formation processes are often dominated by hydrodynamic interactions, and these can be reproduced by a variety of dynamical models. The essence of mesoscopic simulation methods is to combine a significant reduction of degrees of freedom with an artificial dynamical model that reproduces the main dynamical features on long time scales. This results in greatly reduced computation times compared to atomistic simulations.

In this chapter, we focus on the problem of molecular transport in microfluidic geometries, with particular emphasis on transport driven by electric fields. Due to the emerging technology of microfluidic devices like micro-arrays and microchips and their applications

for polymer separation and micromanipulation, the fluid mechanics on these micro- and nano length scales has been an extensive subject of research over the last years. Mesoscopic simulation methods provide useful tools for understanding basic physical mechanisms in these confined systems, and for designing optimal microfluidic setups for given applications.

We intend to give a pedagogical introduction, focusing on the fundamentals of modelling microchannel flow phenomena using coarse-grained techniques. We briefly present the theory behind the most important mechanisms and discuss practical applications. Popular mesoscale simulation methods will be introduced and some examples will be given that illustrate the use and reliability of the presented methods. For further reading we refer to recent reviews in the field such as (Pagonabarraga et al. (2010); Slater et al. (2009); Sparreboom et al. (2010); Viovy (2000)).

## 2. Physics of flow in micro- and nanochannels: A brief introduction

In the following, we will give a brief overview over important factors that govern flow and transport phenomena on the nanoscale. Nanochannels are characterized by an exceedingly large surface-to-volume ratio. Therefore, *boundary effects* become important and dominate the flow properties to a large extent. A second factor that strongly influences the transport properties of molecules in flow is the presence of fluid-mediated *hydrodynamic interactions*. They in turn depend on the nature of the driving force (electric field vs. pressure driven transport or sedimentation) and are affected by the confinement in the channel. Here we can only touch on these issues. For further information we refer to the references.

### 2.1 Hydrodynamics

Throughout this chapter, we will restrict the discussion to Newtonian fluids, *i.e.*, isotropic fluids with a linear stress-strain relationship such as water. Flows in micro- and nanochannels are characterized by low Reynolds numbers. Even though driven flows can reach fluid velocities of meters per second in extreme cases and shear rates as high as  $10^6\text{s}^{-1}$  (Kuang & Wang (2010)), typical values of the Reynolds numbers are still much smaller than unity due to the tiny characteristic length scales. Therefore, it is almost impossible to enter the non-laminar regime, and the hydrodynamic behavior of incompressible Newtonian fluids follows the Stokes equation,

$$\eta \Delta \vec{u} = -\rho \vec{f}^{ext} + \vec{\nabla} P, \quad (1)$$

where  $\vec{u}$  is the local velocity of fluid,  $\vec{f}^{ext}$  an external body force,  $P$  the local pressure, and  $\eta$  the dynamic viscosity of the fluid. We will now discuss the two main factors that determine transport in microchannels in this regime, namely (i) the properties of the channel boundaries and (ii) the hydrodynamic interactions mediated by the fluid.

#### 2.1.1 Hydrodynamic boundary conditions

Due to the large surface-to-volume ratio, flows in microfluidic channels strongly depend on the interaction between the fluid and the boundaries. Surface properties such as the local roughness or wettability can significantly influence the shape and amplitude of flow profiles. On the mesoscale, such effects can often be incorporated in an effective hydrodynamic boundary condition. The most popular is the so-called 'stick' or 'no-slip' boundary condition,

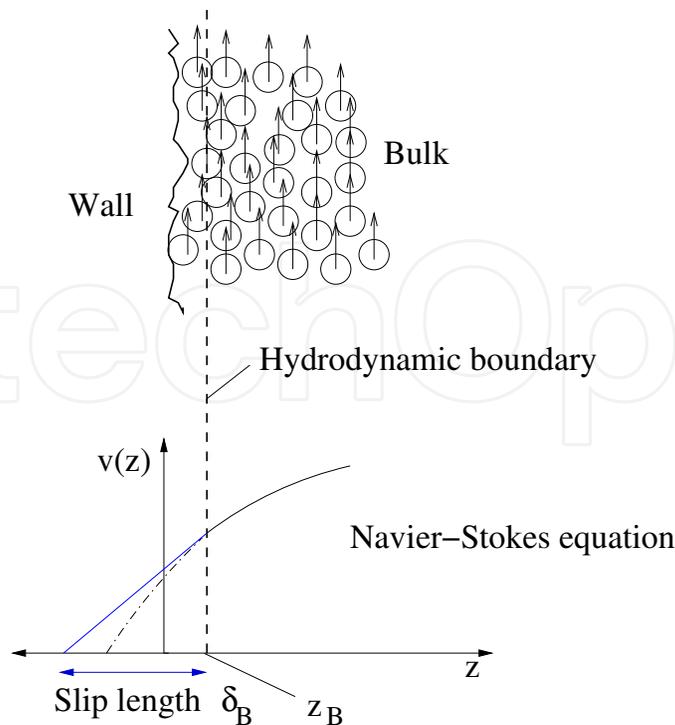


Fig. 1. **Top:** Coarse-grained schematic illustration of fluid flow in close vicinity to the boundaries. The hydrodynamic boundary is not identical to the fluid-solid interface due to atomistic roughness. **Bottom:** Flow profile in the direction normal to the hydrodynamic boundary. In close vicinity to the boundary, the Navier-Stokes equations are not valid. An extrapolation of the flow profile determines the slip length.

where the relative flow velocity is taken to vanish at the boundary (Landau & Lifshitz (1987)),

$$\vec{u}(z)|_{z=z_B} = 0 \tag{2}$$

The coordinate  $z$  is normal to the surface. Due to the atomistic roughness of the boundary, the location of the boundary is not obvious. It constitutes an effective parameter in the boundary condition (2).

The no-slip boundary condition is suitable for describing most macroscopic flows, but it is too restrictive on the nanoscale. Since the friction between the solid surface atoms and the fluid particles is finite, one expects some slippage at the physical boundaries, *i.e.*, the fluid velocity will not strictly vanish. This has been confirmed by a number of experiments on different surfaces (Neto et al. (2005); Pit et al. (2000); Tretheway & Meinhart (2002)). The amount of slippage can be estimated by considering the balance of stresses at the surface. (Kennard (1938)): Let  $x$  be the direction of flow and  $z$  the direction normal to the flow. The viscous stress  $\sigma_{xz}^s$  (Kennard (1938); Landau & Lifshitz (1987)) exerted by microscopic friction between the fluid particles and the solid channel walls

$$\sigma_{xz}^s = \zeta_s u_x(z) \tag{3}$$

is proportional to an *a priori* unknown boundary friction coefficient  $\zeta_s$  and the fluid velocity  $u_x(z)$  relative to the velocity of the wall. The complementary viscous stress generated within

the fluid itself,  $\sigma_{xz}^f$ , is approximated by the corresponding expression in the bulk

$$\sigma_{xz}^f = \eta \frac{\partial}{\partial z} u_x(z). \quad (4)$$

The stability condition (Landau & Lifshitz (1987)) requires that these two stresses are equal, *i.e.*,  $\sigma_{xz}^f|_{z=z_B} = \sigma_{xz}^s|_{z=z_B}$  at the hydrodynamic boundary  $z_B$ . After combining Eqns. (3) and (4) one obtains an effective boundary condition

$$u_x(z)|_{z=z_B} = \delta_B \left( \frac{\partial v_x(z)}{\partial z} \right)_{z=z_B}, \quad (5)$$

which depends on two effective parameters, the hydrodynamic boundary  $z_B$  and the *slip length*

$$\delta_B = \frac{\eta}{\zeta_B}. \quad (6)$$

This so-called *partial-slip boundary condition* (Eqn. (5)) represents the most general hydrodynamic boundary condition on isotropic surfaces which is consistent with all requirements of symmetry. It includes the no-slip boundary condition as the special case  $\delta_B = 0$ . Eq. (6) suggests that the no-slip limit can only be reached for vanishing shear viscosity or infinite wall friction coefficient. It should however be noted that the above "derivation" of (6) relied on the assumption that the Stokes equation is valid directly at the boundary, which is not obvious on rough surfaces. Thus Eq. (6) should not be taken too literally: In practice, slip lengths may become zero or even negative. As long as the actual flow velocities at the physical boundaries stay positive, this is not unphysical. A geometrical interpretation of the slip boundary condition is illustrated in Fig. 1: The slip length is the distance between the hydrodynamic boundary position  $z_B$  and the point where the tangent to the Stokes flow profile at  $z_B$  hits zero.

### 2.1.2 Hydrodynamic interactions

Next we discuss the transport of molecules or nanoparticles in a fluid under the influence of external forces. Consider a particle or a system of particles made of several idealized point-like "monomers". If a force acts on a monomers, it will respond by moving and drag the surrounding fluid along. This generates a flow field and convective motion at the positions of the other monomers. Thus *all* monomers will respond to a force  $\vec{f}_i$  acting on one monomer  $i$ . This fluid-mediated nonlocal dynamical response is termed hydrodynamic interaction (Dhont (1996)). Specifically, the force  $\vec{f}_i$  on a pointlike particle at position  $\vec{r}_i$  excites a convective flow with velocity

$$\vec{u}(\vec{r}_j) = \mathbf{T}(\vec{r}_j, \vec{r}_i) \vec{f}_i \quad (7)$$

at the position  $\vec{r}_j$  of the pointlike particle  $j$ , which is proportional to  $\vec{f}_i$  with a tensorial coupling function  $\mathbf{T}$ , the 'Oseen tensor'. Since the Stokes equation is linear, all these flow velocities add up linearly. The Oseen tensor depends on the boundary conditions. In the bulk, it is a function of the difference  $\vec{r} = \vec{r}_i - \vec{r}_j$  only and given by

$$T_{\alpha\beta}(\vec{r}) = \frac{1}{8\pi\eta r} \left( \delta_{\alpha\beta} + \hat{r}_\alpha \hat{r}_\beta \right). \quad (8)$$

with  $\hat{r} = \vec{r}/r$ . This corresponds to a long-range "interaction" which decays like  $1/r$ .

Due to their long-range nature, the hydrodynamic interactions strongly influence the dynamics and transport properties of single molecules and molecule assemblies. For example, polymer coils in solution move like a Stokes sphere due to hydrodynamic interactions, with an effective friction coefficient  $\zeta$  which is proportional to their radius ("Zimm dynamics"), whereas  $\zeta$  is simply proportional to the chain length in the absence of hydrodynamics ("Rouse dynamics") (Doi & Edwards (1986)).

In nanochannels, the presence of boundaries further complicates the situation. Since the boundaries break translational symmetry, the assumption  $\mathbf{T}(\vec{r}_i, \vec{r}_j) = \mathbf{T}(\vec{r}_i - \vec{r}_j)$  is no longer justified and the Oseen tensor depends on both arguments,  $\vec{r}_i$  and  $\vec{r}_j$ , independently. In homogeneous channels, one may still assume  $\mathbf{T}(\vec{r}_i, \vec{r}_j) = \mathbf{T}(\vec{r}_{i\parallel} - \vec{r}_{j\parallel}; \vec{r}_{i\perp}, \vec{r}_{j\perp})$ , where  $\vec{r}_{\parallel}$  is the component of  $\vec{r}$  along the channel and  $\vec{r}_{\perp} = \vec{r} - \vec{r}_{\parallel}$ . In structured channels, even this last symmetry assumption breaks down.

A second important effect of boundaries is that they constrain the flow. This affects the hydrodynamic interactions dramatically. For example, the hydrodynamic interactions between two particles close to a single no-slip wall decay like  $1/r^3$  instead of  $1/r$  (Dufresne et al. (2000)). The hydrodynamic interactions between two particles confined in a thin planar slit with no-slip boundaries decay like  $1/r^2$ , *i.e.*, much faster than in the bulk, albeit still long-range. (Diamant et al. (2005); Liron & Mochon (1976)). Even these remaining long-range contribution may effectively average out in many applications, since the two-dimensional Oseen tensor in slits is traceless and its orientational average vanishes. It has been argued that polymers confined in thin sheets should therefore exhibit Rouse-like dynamics (Tlusty (2006)). In quasi one dimensional thin channels with stick boundaries, hydrodynamic interactions are altogether screened on distances larger than the channel width (Cui et al. (2002)).

## 2.2 Electrokinetics

If external electric fields are involved, flow and transport processes in microchannels are governed by the interplay of hydrodynamic and electrostatic driving forces. Both the hydrodynamic and the electrostatic interactions are long-range, thus none of the two necessarily dominates over the other. We consider the transport of charged macromolecules or colloids ("macroions") in solution due to electric fields. On the one hand, typical buffers used in (bio)technological applications have high ionic strengths (*i.e.*, high salt concentrations), such that macrocharges are efficiently screened with Debye screening lengths in the nanometer range. On the other hand, the (micro-)ions in solution respond to external fields just like the macroions. Furthermore, external fields may induce flow in microchannels by means of an effect called electroosmosis, which also contributes to the effective transport.

### 2.2.1 Basic concepts

In the following we will take the correlations between microions to be negligible. Correlation effects become important at high surface charges, low temperatures, and/or if microions have a high valency (Boroudjerdi et al. (2005)). In most practical applications, they can be neglected and one may apply mean-field approaches. At thermal equilibrium, the mean-field

theory of choice is the *Poisson-Boltzmann* theory (Hunter (1989); Israelachvili (1991); Lyklema (1995); Viovy (2000)), which describes the distribution of the microions by a combination of the Poisson equation and the Boltzmann distribution. The theory neglects the excluded-volume of the counterions and regards them as pure uncorrelated point-like particles. Ions of type  $i$  are distributed in space according to  $\rho_i(\vec{r}) \propto \exp(-Z_i e \psi(\vec{r})/k_B T)$ , where  $Z_i$  is the valency of the ion  $i$ ,  $e$  the elementary charge,  $k_B T$  the thermal energy, and  $\psi(\vec{r})$  the mean electrostatic potential at position  $\vec{r}$ . Combining this with the Poisson equation, one obtains the self-consistent equation

$$\epsilon_r \Delta \psi(\vec{r}) = -\rho_{\text{macro}}(\vec{r}) - \sum_i Z_i e \rho_i(\vec{r}) = -\rho_{\text{macro}}(\vec{r}) - \sum_i Z_i e \rho_{0,i} e^{-\frac{Z_i e \psi(\vec{r})}{k_B T}} \quad (9)$$

with the dielectric constant  $\epsilon_r$ , the macroion charge density  $\rho_{\text{macro}}(\vec{r})$ , and the bulk microion densities  $\rho_{0,i}$  (the hypothetical densities at infinite distance from the macroions). Here the  $\rho_{\text{macro}}$  may also refer to a charged surface.

Due to its nonlinear character, the Poisson-Boltzmann equation (9) can usually not be solved in closed form. If the potential  $\psi$  is small, it can be linearized by expanding the exponent. This yields the famous *Debye-Hückel* equation

$$(\Delta - \lambda_D^{-2})\psi(\vec{r}) = -\frac{\rho_{\text{macro}}(\vec{r})}{\epsilon_r} \quad (10)$$

with the Debye-Hückel screening length

$$\lambda_D = \sqrt{\frac{\epsilon_r k_B T}{\sum_i \rho_{0,i} (Z_i e)^2}}. \quad (11)$$

Here we have used  $\sum_i \frac{Z_i e}{\epsilon_r} \rho_{0,i} = 0$ . According to the Debye-Hückel theory, the effective interactions between macroions decay exponentially with the characteristic length  $\lambda_D$ . The range of validity of the Debye-Hückel approximation is very limited, it breaks down already for moderate macroion charges and/or for highly concentrated ion solutions. Nevertheless, the exponential behavior often persists in systems where the Debye-Hückel approximation is not valid. For highly charged macroions, the Debye-Hückel theory can still be applied far from the macroions, but the macroion charge must be replaced by an effective "renormalized charge" (Alexander et al. (1984); Trizac et al. (2002)). For high ion concentrations, detailed theoretical studies have lead to the conclusion that the results of the Debye-Hückel can still be used in a wide parameter range if  $\lambda_D$  is replaced by a modified effective screening length (Kjellander & Mitchell (1994); McBride et al. (1998); Mitchell & Ninham (1978)).

So far, these are equilibrium considerations. At non-equilibrium and in the presence of flow, the time evolution of the ion distributions at the mean-field level are described by a convection-diffusion equation, the *Nernst-Planck* equation

$$\frac{\partial \rho_i}{\partial t} + \nabla \cdot (\rho_i \vec{u}) = \nabla \cdot D_i (\nabla \rho_i + \frac{Z_i e}{k_B T} \rho_i \nabla \psi). \quad (12)$$

Here  $D_i$  are the diffusivities of the ion species  $i$ , and  $\vec{u}(\vec{r})$  is the local fluid velocity. The Nernst-Planck equation, the Stokes (or Navier-Stokes) equation and the Poisson equation

constitute a complete set of equations that fully determine the dynamics of ionic fluids. In stationary systems without flows, they are solved by the Poisson-Boltzmann distribution.

### 2.2.2 Boundary layers and electroosmotic flow

In contact with a solvent, many materials acquire charges due to the ionization or dissociation of surface groups or the adsorption of ions from solution (Israelachvili (1991)). Take for example PDMS (polydimethylsiloxane), a material commonly used for the fabrication of microfluidic devices. At PDMS/water interfaces,  $(\text{OH})^-$  groups dissociate at the end of the PDMS chains. As a result, the walls of PDMS channels become charged and an ion layer builds up in front of the surfaces to screens their charge: An *electrical double layer* forms.

The structure of the ion layer is commonly described by an effective picture that distinguishes between two regions: In the outer region, the *diffuse layer*, the ions are mobile, they can thermally be removed and act as if they are part of the solution. Under the influence of force, they will drag the fluid along. The inner region, the *Stern layer*, includes ions which stick to the surface. The two regions are taken to be separated by a "plane of shear" for the fluid flow. In the coarse-grained mean-field picture adopted in this chapter, the motion of the ions in the diffuse layer is described by the Nernst-Planck equation (12), the plane of shear corresponds to the hydrodynamic boundary, and the complex of solid surface and Stern layer is replaced by an effective boundary condition which is characterized by an effective surface charge and an effective slip length.

At equilibrium, the distribution of ions in the diffuse layer is determined by the Poisson-Boltzmann equation. In the case of symmetric planar slit channels confined by homogeneous surfaces, and in the absence of salt (*i.e.*, the fluid contains only the counterions that neutralize the surface charges), the problem can be solved analytically (Israelachvili (1991)). The potential in the direction  $z$  perpendicular to the surface is then given by

$$\psi(z) = \frac{k_B T}{Ze} \log(\cos^2(\kappa z)) \quad (13)$$

with the screening constant

$$\kappa^2 = \frac{(Ze)^2 \rho_0}{2\epsilon_r k_B T}. \quad (14)$$

Here  $\rho_0$  denotes the counterion density in the middle of the channel, and the  $z$ -direction is perpendicular to the channel. In the presence of salt an approximate solution can be obtained within the Debye-Hückel approximation, giving

$$\psi(z) \propto (e^{z/\lambda_D} + e^{-z/\lambda_D} - 2), \quad (15)$$

which results in an exponential decay of the ion distribution close to the boundaries with the decay length  $\lambda_D$ , the Debye-Hückel screening length  $\lambda_D$  from Eq. (11). The equilibrium results (13) and (15) remain valid in stationary situations with constant flow parallel to the surface, since the parallel and the perpendicular directions in Eq. (12) then decouple. In such cases, they describe the  $z$ -dependent part of the potential  $\psi$ .

If an external electric field is applied to the solution, it generates a net volume force in the diffuse ion layer, which induces flow, the 'electroosmotic flow' (EOF) (Hunter (1989)).

This phenomenon provides one with a convenient tool to manipulate and control flows in microchannels. Furthermore, it significantly influences the effective transport of nanoobjects in such channels.

The EOF flow profiles can be calculated by combining the Stokes equation (Eq. (1)) with the Poisson-Boltzmann theory. For simplicity, we consider again our symmetric slit channel. The net charge density in the channel,  $\rho(z) = \sum_i (Z_i e) \rho_i(z)$ , generates a force density  $f_{\parallel}(z) = \rho(z) E_{\parallel}$  in the fluid, where  $E_{\parallel}$  is a parallel external electric field. Comparing the Poisson equation for the electrostatic potential,

$$\frac{\partial^2 \psi(z)}{\partial z^2} = -\frac{\rho(z)}{\epsilon_r}, \quad (16)$$

with the Stokes equation for the parallel component of the fluid velocity,

$$\eta \frac{\partial^2 u_{\parallel}(z)}{\partial z^2} = -f_{\parallel}(z) = -\rho(z) E_{\parallel}, \quad (17)$$

one finds immediately

$$\frac{\partial^2}{\partial z^2} u_{\parallel}(z) = \frac{\epsilon_r E_{\parallel}}{\eta} \frac{\partial^2}{\partial z^2} \psi(z). \quad (18)$$

In a symmetric channel, the profiles  $u_{\parallel}(z)$  and  $\psi$  satisfy the boundary condition  $\partial_z u_{\parallel}|_{z=0} = \partial_z \psi|_{z=0} = 0$  at the center of the channel. This gives the relation (Smiatek & Schmid (2010))

$$u_{\parallel}(z) = \frac{\epsilon_r E_{\parallel}}{\eta} (\psi(z) - \psi(0)) + u_{\text{EOF}} \quad (19)$$

The constant  $u_{\text{EOF}} = u_{\parallel}(0)$  in Eq. (19) is determined by the hydrodynamic boundary conditions at the surface of the channel. For example, the counterion-induced electroosmotic flow in the presence of slip resulting from Eqs. (19) and (13) is given by (Smiatek et al. (2009))

$$u_{\parallel}(z) = \frac{\epsilon_r k_B T}{Ze\eta} E_{\parallel} \left( \log(\cos^2(\kappa z)) - \log(\cos^2(\kappa z_B)) - 2\kappa \delta_B \tan(\kappa z_B) \right), \quad (20)$$

where  $\delta_B$  is the slip length and  $z_B$  the position of the hydrodynamic boundary (cf. Eq. (6)). In the presence of salt, the EOF profiles vary mostly at the boundaries and the velocity profile saturates in the center of the channel (plug flow). In that case, the constant  $u_{\text{EOF}}$  can be identified with the EOF velocity. Setting  $\psi(0) = 0$ , Eq. (19) combined with the partial-slip boundary condition (5) results in the general simple expression (Bouzigues et al. (2008); Smiatek & Schmid (2010))

$$u_{\text{EOF}} = -\frac{\epsilon_r \psi_B}{\eta} (1 + \kappa_s \delta_B) E_{\parallel} =: -\frac{\epsilon_r \psi_{\zeta}}{\eta} E_{\parallel}. \quad (21)$$

where  $\psi_B$  is the electrostatic potential at the hydrodynamic boundary and we have defined a 'surface screening parameter'

$$\kappa_s := \mp \frac{\partial_z \psi}{\psi} \Big|_{z=\mp z_B}. \quad (22)$$

In the Debye-Hückel regime,  $\kappa_s$  is simply the inverse Debye-Hückel length (Joly et al. (2004)). More generally, the relation (21) also remains valid in the regime where nonlinear effects become important. On no-slip surfaces, Eq. (21) reduces to the well-known Smoluchowski result for electroosmotic flow (Hunter (1989)). In the presence of slippage, the amplitude of the induced flow can be enhanced significantly. This is also found experimentally (Bouzigues et al. (2008)). The combined effect of slip, surface potential, and surface screening parameter can be incorporated into one single parameter  $\psi_\zeta = \psi_B(1 + \kappa_s \delta_B)$ , the so-called "Zeta-potential", which fully characterizes the electroosmotic response of the surface to an applied electric field. (Hunter (1989); Lyklema (1995))

So far, we have considered simple slit channels. However, our results can also be used to assess EOF profiles in more complex channel geometries. We assume that the diffuse layer is thin compared to the channel dimensions. Then, we make two important observations: First, Eq. (21) still can be applied close to the boundaries and gives an effective *boundary condition* for a steady-state velocity field  $\vec{u}_{\text{EOF}}(\vec{r})$  inside the channel, which depends linearly on the applied external field  $\vec{E}^{(ext)}$ ,

$$\vec{u}_{\text{EOF}} = \mu_{\text{EOF}} \vec{E}^{(ext)} \quad \text{with} \quad \mu_{\text{EOF}} = -\frac{\epsilon_r \psi_\zeta}{\eta}. \quad (23)$$

(In a steady-state situation, the external field in the vicinity of a wall is necessarily parallel to the wall.) Cummings and coworkers have shown that for laminar incompressible flow, the boundary condition (23) effectively defines the flow inside the channel (Cummings et al. (2000)). Second, the EOF flow field inside the channel obeys the Laplace equation,  $\Delta \vec{u}_{\text{EOF}} = 0$ . The same holds for the external electric field  $\vec{E}^{(ext)}$ . Provided all channel walls are made of the same material (same slip length, same surface potential  $\psi_B$ ) and provided Eq. (23) also holds at the inlet and outlet boundaries of the channel, Eq. (23) thus becomes valid everywhere in the channel. Due to this remarkable 'similitude' property, the EOF velocity profile in such channels is simply proportional to the applied electric field in the channel, which can be calculated by solving the Laplace equation in the channels with von-Neumann boundary conditions. The proportionality factor is  $\mu_{\text{EOF}}$ , the so-called 'electroosmotic mobility'.

We note that the condition on the inlet and outlet is less restrictive than it may seem: One can always separate a steady-state velocity field  $\vec{u}(\vec{r})$  into an EOF component  $\vec{u}_{\text{EOF}}(\vec{r})$  which satisfies the EOF boundary conditions, and a residual component with no-slip boundary conditions at the channel walls and arbitrary boundary conditions at the inlet and outlet.

### 2.2.3 Electrophoresis

Last in this section we discuss the transport of charged nanoobjects (*i.e.*, colloids or polyelectrolytes) in microchannels under the influence of constant electric fields. In the linear regime, the velocity of the nanoobject is proportional to the field,

$$\vec{v}_P = \mu_t \vec{E}, \quad (24)$$

The proportionality constant  $\mu_t$  denotes the total electrophoretic mobility and results from two contributions (Smiatek & Schmid (2010); Streek et al. (2005)): The bare electrophoretic mobility  $\mu_e$  in a fluid at rest and the background EOF velocity  $\vec{u}_{\text{EOF}}$ . Due to the similitude discussed above between the steady-state velocity field of an EOF and the external field  $\vec{E}$ , the EOF flow field can be expressed in terms of an electroosmotic mobility  $\mu_{\text{EOF}}$  (Eq. (23)), and the

total electrophoretic mobility of the nanoparticle can be written as

$$\mu_t = \mu_{EOF} + \mu_e. \quad (25)$$

The bare electrophoretic mobility  $\mu_e$  subsumes the intrinsic response of the nanoobject to the external field. For large colloids with a thin Debye layer, this response is dominated by the same mechanisms than the EOF at the walls: The electric field induces EOF at the surface of the colloid, which results in a velocity difference (23) between the colloid and the surrounding fluid. Thus  $\mu_e$  is again given by the Smoluchowsky result in this case,

$$\mu_e = \epsilon_r \psi_\zeta / \eta, \quad (26)$$

where the Zeta-potential  $\psi_\zeta$  characterizes the surface of the colloid, *i.e.*, it incorporates the effects of the Stern layer as well as possible slippage at the hydrodynamic boundary.

For small nanosized particles or ions, the situation is more complicated. The effective friction of a charged particle in an electric field mainly results from the combination of three factors: (i) The direct friction of the particle with the surrounding fluid, which can be described by a bare mobility  $\mu^0$ . It determines the mobility of the nanoparticle in the absence of any other ions. (ii) The solvent-mediated friction of the particle with the surrounding cloud of oppositely charged ions (the "ionic atmosphere"). They are dragged in the opposite direction by the external field, which generates additional friction. This effect is called "electrophoretic retardation". (iii) The distortion of the ion cloud in the electric field. Since the cloud is a dynamic phenomenon, ions move in and out, it cannot adjust instantaneously to the motion of the nanoparticle. Moreover, the electric field drives the nanoparticle and the cloud apart. The cloud is somewhat behind the nanoparticle and pulls it back, thus slowing it down. This is called the "relaxation effect". The interplay of these three factors and their effect on the electrophoretic mobility of ions has been discussed already in 1926 by Onsager (Onsager (1926; 1927)). He predicted that the bare mobility of ions  $\mu^0$  in electrolytes should be renormalized according to

$$\mu_e = \mu^0 - (A\mu^0 + B) \sqrt{\sum_j \rho_j Z_j^2}, \quad (27)$$

where the factors  $A$  and  $B$  represent the relaxation effect and the electrophoretic effect. Explicit expressions for  $A$  and  $B$  for electrolytes having any number of ionic species were derived by Onsager and Fuoss in (Onsager & Fuoss (1932)). More details can be found in (Wright (2007)).

In systems of several nanoparticles or of extended molecules, hydrodynamic interactions become important. As discussed in Section 2.1.2, they can crucially affect the dynamics and transport properties in external fields. In the case of electrophoresis, one must not only consider the direct hydrodynamic interactions between the macroions, but also the effect of the surrounding microion cloud. In an external electric field, the polyelectrolyte and its surrounding ion cloud move in opposite directions, and the net momentum transferred to the solvent adds up to zero. As a result, the hydrodynamic interactions are largely screened (Manning (1981)). Within the Debye-Hückel approximation, the corresponding modified Greens function can be calculated analytically. The convective flow created by a constant electric field  $\vec{E}$  acting on a particle at position  $\vec{r}_i$  is given by (Barrat & Joanny (1996);

Long & Ajdari (2001))

$$\vec{u}(\vec{r}_j) = \mathbf{G}(\vec{r}_j, \vec{r}_i) \vec{E} \quad (28)$$

with a Greens function that depends on the Debye screening length  $\lambda_D$  via

$$G_{\alpha\beta}(\vec{r}) = \frac{e^{-r/\lambda_D}}{4\pi\eta r} \left( \frac{2}{3}\delta_{\alpha\beta} + \frac{\lambda_D^2}{r^2} \left( 1 + \frac{r}{\lambda_D} + \frac{r^2}{3\lambda_D^2} - e^{r/\lambda_D} \right) (\delta_{\alpha\beta} - 3\hat{r}_\alpha \hat{r}_\beta) \right) \quad (29)$$

These expressions take the place of Eqs. (7) and (8) in electrophoresis. The modified Greens function  $\mathbf{G}$  contains an exponentially decaying part, and an algebraic contribution which decays with  $1/r^3$  (instead of  $1/r$ ) and vanishes upon averaging over all directions. Thus, the effect of hydrodynamic interactions is greatly reduced. We note that this only applies for the response of charged particles to a constant electric field. The *diffusive* behavior of charged particles is still governed by the usual, long-range hydrodynamic interactions.

The remarkable electrohydrodynamic screening effect in electrophoresis discussed above has important practical consequences. A common task in bioanalytics is length-dependent separation and sorting of charged polymers, *e.g.*, DNA fragments. Strategies to sort molecules in micro- or nanofluidic channels are highly desirable as cheap and versatile alternative to standard gel electrophoresis. In the absence of hydrodynamic interactions, however, the electrophoretic mobility of polyelectrolytes does not depend on the chain length. Thus electric fields cannot be used to separate polymers in simple straight channel geometries, and one has to resort to more sophisticated approaches.

### 3. Mesoscale simulation methods

After this brief overview over physical mechanisms governing flow in microchannels and the transport of nanoobjects on mesoscopic scales, we will now introduce different methods to study these phenomena by computer simulations. We focus on length scales which are large compared to the solvent particles (water), but small enough that thermal fluctuations are important and nanoobjects can be resolved and traced individually. On even smaller length scales, the method of choice is atomistic Molecular Dynamics. It is used, *e.g.*, to study the microscopic factors that determine surface slip on specific surfaces (Bocquet & Barrat (1994); Joly et al. (2004; 2006)). On much larger length scales, deterministic approaches can be used and one may resort to the numerical methods that have been developed in the context of computational fluid dynamics to solve the Navier-Stokes equations in various geometries. A software package for studying electrohydrodynamic phenomena in colloidal suspensions at this level is freely available at (Kim et al. (2006))

<http://www-tph.cheme.kyoto-u.ac.jp/kapsel/>.

Electrokinetic simulations are challenging, because one has to deal with two types of long-range interactions simultaneously: Long-range hydrodynamics and long-range electrostatics. In the following, we will first discuss mesoscale methods for simulating hydrodynamic phenomena in nanoparticle suspensions. Then we will present a selection of simulation methods for studying electrokinetic phenomena.

### 3.1 Simulation methods for hydrodynamic flows

In mesoscale simulations, one is typically not interested in the explicit dynamical behavior of the solvent molecules, as typical time scales are much smaller than the time scales of interest. For example, the relaxation time of water is 15 GHz, whereas the fastest time scale for reorganization of ionic layers in aqueous solutions is in the MHz regime (ionic migration in the double layer sets in at 1-10 MHz). A coarse-grained simulation model must therefore mainly capture the hydrodynamics of the solvent. To make the simulations more efficient, the molecular solvent can thus be replaced by something simpler. In the case of Newtonian fluids, such simplified model solvents must fulfill three main requirements:

- It must reproduce the Navier-Stokes equation. This means that it is a fluid (*i.e.*, it yields to stress), that it is "simple" (no long-range interactions, no memory effects, no internal order, no phase transitions), and that it preserves momentum locally.
- Since fluctuations are important, it should have a temperature. Simulations of microfluidic transport are typically nonequilibrium simulations where energy is constantly fed into the system. Therefore, algorithms should be available that remove this energy from the system, *i.e.*, coupling to a thermostat should be possible.
- Strategies to implement arbitrary hydrodynamic boundary conditions and interactions with nanoparticles should exist.

Other symmetry properties such as isotropy of space or translational invariance would also be desirable. However, they are not strictly mandatory. We shall see that some of the most successful models do not meet these symmetry requirements.

The reader may have noted that the requirements sketched above are already met by ideal gases. In fact, many mesoscopic fluid models in use today are actually ideal gas models. This implies that the fluid is not incompressible, which is acceptable as long as typical velocities  $V$  are much smaller than the speed of sound  $c$  in the model. Ideal gas models can be applied to study incompressible fluid dynamics if the *Mach number*  $M = V/c$  is small.

#### 3.1.1 Dissipative Particle Dynamics (DPD)

Among the mesoscopic simulation methods for fluids, the DPD method comes closest to classical Molecular Dynamics (MD). It was originally developed as a method for fluid simulations that combines elements of Lattice Gas Automata and MD (Hoogerbrugge & Koelman (1992); Koelman & Hoogerbrugge (1993)). In its modern version (Español & Warren (1995)), it is basically a momentum-conserving, Galilean invariant thermostat which creates a well-defined canonical ensemble. In DPD simulations, particles  $i$  obey Newton's equation of motion with the forces

$$\vec{F}_i^{DPD} = \vec{F}_i^C + \sum_{i \neq j} (\vec{F}_{ij}^D + \vec{F}_{ij}^R), \quad (30)$$

where  $\vec{F}_i^C$  are the regular conservative forces acting on  $i$ , and  $\vec{F}_{ij}^D, \vec{F}_{ij}^R$  are pairwise dissipative and stochastic forces. The dissipative forces  $\vec{F}_{ij}^D$  have the form

$$\vec{F}_{ij}^D = -\gamma_{DPD} \omega_{DPD}(r_{ij})(\hat{r}_{ij} \cdot \vec{v}_{ij}) \hat{r}_{ij} \quad (31)$$

with the friction coefficient  $\gamma_{\text{DPD}}$ , and depend on the interparticle distance  $r_{ij}$  and the unit vector  $\hat{r}_{ij}$  connecting the two particles  $i$  and  $j$ . The random force  $\vec{F}_{ij}^R$  is chosen such that it fulfills the fluctuation-dissipation relation

$$\vec{F}_{ij}^R = \sqrt{2\gamma_{\text{DPD}}k_B T \omega_{\text{DPD}}(r_{ij})} \theta_{ij} \hat{r}_{ij}, \quad (32)$$

with the Boltzmann constant  $k_B$  and the temperature  $T$ , where  $\theta_{ij} = \theta_{ji}$  is a Gaussian distributed random number with zero mean and unit variance. The weighting function  $\omega_{\text{DPD}}$  is arbitrary and often chosen as a linearly decaying function with a cutoff  $r_c$ ,

$$\omega_{\text{DPD}}(r_{ij}) = \begin{cases} 1 - \frac{r_{ij}}{r_c} & : r_{ij} < r_c \\ 0 & : r_{ij} \geq r_c \end{cases} \quad (33)$$

Eq. (30) can be integrated by an ordinary MD integration scheme like the Velocity-Verlet algorithm (Frenkel & Smit (2001)).

In DPD simulations, arbitrary equation of states can be implemented through appropriate conservative interactions  $\vec{F}_i^C$ . In practice, the conservative potentials are often chosen pairwise and soft and with linearly decaying forces (Groot & Warren (1997)). When studying fluid flow, one may also simply turn them off. For such an "ideal gas" DPD fluid, approximate expressions for the viscosity have been derived *via* a Fokker-Planck formalism (Marsh et al. (1997)). The shear viscosity of the fluid can be tuned by varying the density, the friction parameter  $\gamma_{\text{DPD}}$  or the cutoff parameter of the dissipative interactions,  $r_c$ . Since the DPD method was first introduced by Hoogerbrugge and Koelman, various variants and extensions have been proposed. A recent review can be found in (Pivkin et al. (2010)).

Compared to other mesoscale simulation methods, DPD simulations are relatively expensive because they require the evaluation of pairwise forces in every time step. Nevertheless, they present many advantages due to the fact that they are so similar to standard MD simulations. They can easily be mixed with MD schemes, to the point that the viscous DPD interaction can simply be used to thermostat MD simulations (Soddemann et al. (2003)). Interactions with nanoparticles can be implemented in a straightforward manner through the conservative interactions. Arbitrary hydrodynamic boundary conditions with slip lengths ranging from full slip to no-slip or even slightly negative can be implemented using a recently developed "tunable slip" algorithm (Smiatek et al. (2008)). This algorithm introduces a DPD-like friction interaction with the wall. Slip lengths can be adjusted by varying the amplitude of the friction force or the interaction range, and an approximate expression for the resulting slip length can be derived analytically which is in excellent agreement with simulations. The method can also be used to implement an EOF boundary condition with fixed surface velocity  $\vec{u}_{\text{EOF}}$  (by choosing a no-slip boundary condition on a hypothetically moving boundary). Alternative methods for implementing no-slip boundaries in DPD simulations can be found, *e.g.*, in Refs. (Revena et al. (1998; 1999)).

### 3.1.2 Multiparticle collision dynamics (MPCD)

One drawback of the DPD method described in the previous section is that DPD fluids typically have low Schmidt numbers. The dimensionless Schmidt number  $Sc = \eta/\rho D$

measures the ratio of momentum diffusivity (viscosity  $\eta$ ) and mass diffusivity (diffusion constant  $D$ ). In gases, particle transport dominates and Schmidt numbers are typically of order unity. In liquids, dissipative transport dominates and typical Schmidt numbers are high. For example, the Schmidt number of water is  $Sc \sim 700$ . In DPD fluids, typical Schmidt values are of order unity like in gases, hence the momentum transport is mostly coupled to particle transport. High, fluid-like Schmidt numbers are not accessible in DPD simulations.

This problem can be overcome in the MPCD method, which is another increasingly popular method for simulating fluid flow. MPCD, also known as stochastic rotation dynamics (SRD) is a particle-based approach for simulating fluid flows which is motivated by the Boltzmann equation for transport in gases (Malevanets & Kapral (1999; 2000)). An extensive recent review can be found in (Gompper et al. (2009)).

The basic algorithm is an alternating sequence of streaming and a collision steps. In the streaming step the positions of the particles  $i$  are updated *via*

$$\vec{r}_i(t + \Delta t) = \vec{r}_i(t) + \Delta t \vec{v}_i(t). \quad (34)$$

In the collision step, the particles are sorted into cells (typically cubic with cell size  $a$  chosen such that one has about 3-20 particles per cell), and the center-of mass velocity  $\vec{u}$  is evaluated in each cell. The collision update then reads

$$\vec{v}_i(t + \Delta t) = \vec{u} + \mathcal{R} \delta \vec{v}_i(t) \quad (35)$$

where  $\mathcal{R}$  is a collision operator acting on the local velocity  $\delta \vec{v}_i(t) = \vec{v}_i(t) - \vec{u}$ . In the classical MPCD scheme,  $\mathcal{R}$  is a rotation by a fixed angle  $\alpha$  about a random axis. Malevanets and Kapral have shown that this scheme generates a Maxwellian velocity distribution at equilibrium and reproduces the hydrodynamic equations with an ideal gas equation of state (Malevanets & Kapral (1999)). The shear viscosity can be calculated analytically, *e.g.*, *via* a Green-Kubo formalism, and it can be tuned through tuning  $a$ ,  $\Delta t$ , and/or the rotation angle  $\alpha$ . In addition, one can also tune the Schmidt number by adjusting the time step  $\Delta t$ . Large fluid-like Schmidt numbers can be realized by choosing small time steps.

Several variants of MPCD have been proposed. The classical version of the MPCD algorithm is not Galilean invariant due to the underlying grid structure. While this is often acceptable, it may cause problems if the mean free path of particles is smaller than the cell size. Galilean invariance can be restored by performing a random shift of the cells before each collision step (Ihle & Kroll (2003)). Another issue is isotropy of space. The classical MPCD scheme does not strictly conserve angular momentum. This can be fixed by choosing a stochastic collision step, where new relative velocities  $\delta \vec{v}_i$  are generated randomly from a Maxwell distribution subject to additional constraints (Noguchi et al. (2007)). The stochastic variant also provides one with a natural implementation of a thermostat, which is necessary for nonequilibrium simulations.

Since MPCD is a particle-based method, interactions with nanoparticles can be implemented in a straightforward manner, just like in DPD simulations. Likewise, the tunable slip method for implementing arbitrary hydrodynamic boundary conditions in DPD simulations (Smiatek et al. (2008)) should also work for MPCD models, but this has not yet been explored. Instead, hydrodynamic boundary conditions in MPCD are usually realized by implementing suitable *reflection rules* on hard surfaces. Full-slip boundaries can be obtained with specular

reflections. No-slip boundaries are obtained with "bounce-back" reflections, where the velocities of particles are reversed when they hit the boundary. Even though these reflections take place during the streaming steps, the collision steps are also affected by the presence of hard walls, due to the fact that the number of MPCD particles is reduced in cells that contain boundaries. Since the local viscosity of the fluid depends on the local density, a local reduction may cause artefacts. One common strategy to overcome this problem is to randomly insert virtual MPCD particles in underfilled cells (Lamura et al. (2001)), with velocities drawn randomly from a Maxwell distribution which is centered about the mean velocity of the solid in no-slip cases, or the fluid in full-slip cases. Based on these realizations of full-slip and no-slip boundaries, one can also implement arbitrary partial-slip boundary conditions by randomly switching between specular and bounce-back reflections in the streaming step, and suitably interpolating the mean velocity of the pseudoparticles in the collision step (Whitmer & Luijten (2010)).

### 3.1.3 Lattice-Boltzmann method (LB)

Similar to MPCD, the LB method realizes a Boltzmann equation, but this time fully on the lattice (Dünweg & Ladd (2009); Succi (2001); Yeomans (2006)). Each lattice site  $\vec{r}$  carries a set of discrete velocity distribution functions  $n_i(\vec{r}, t)$ , which describe the partial density of fluid particles with velocity  $\vec{c}_i$  at time  $t$ . The finite set of velocities  $\{\vec{c}_i\}$  contains only lattice vectors. For example, the velocity set of the popular D3Q19 model contains the zero velocity and the vectors that connect a node with its nearest and next-nearest neighbors on a cubic lattice. The velocity unit is the "lattice velocity"  $c = a/\Delta t$  with the lattice parameter  $a$  and the time step  $\Delta t$ .

As in MPCD, the LB dynamics is a sequence of alternating collision and streaming steps. The distributions are first reshuffled in the collision step,

$$n_i^*(\vec{r}, t) = n_i(\vec{r}, t) + L_{ij}(n_j(\vec{r}, t) - n_j^{\text{eq}}(\rho, \vec{u})) \quad (36)$$

and then evolve in the streaming step according to

$$n_i(\vec{r} + \vec{c}_i \Delta t, t + \Delta t) = n_i^*(\vec{r}, t) \quad (37)$$

The collision step describes particle scattering events which relax the distributions towards a local equilibrium distribution  $\{n_j^{\text{eq}}\}$  with a linear collision operator  $L_{ij}$ . The simplest choice for  $L$  is  $L_{ij} = -\delta_{ij}/\tau$ , corresponding to a lattice Bhatnagar-Gross-Krook operator with a single relaxation time (Bhatnagar et al. (1954)). Multiple relaxation times schemes have been conceived as well and have the advantage that shear and bulk viscosity can be tuned separately (Dünweg & Ladd (2009)). The local pseudo equilibrium distribution  $\{n_i^{\text{eq}}\}$  depends on the local mass density  $\rho(\vec{r}, t)$  and momentum density  $\vec{j}(\vec{r}, t)$ , which are related to the moments of the distribution function through

$$\rho(\vec{r}, t) = \sum_i n_i(\vec{r}, t), \quad \vec{j}(\vec{r}, t) = \rho(\vec{r}, t) \vec{u}(\vec{r}, t) = \sum_i n_i(\vec{r}, t) \vec{c}_i. \quad (38)$$

It is typically constructed as a second order expansion in the velocity,

$$n_i^{\text{eq}}(\rho, \vec{u}) = \rho(A_q + B_q(\vec{c}_i \cdot \vec{u}) + C_q u^2 + D_q(\vec{c}_i \cdot \vec{u})^2) \quad (39)$$

with coefficients  $A_q, B_q, C_q, D_q$  chosen such that  $\{n_i^{\text{eq}}\}$  satisfies the constraints  $\sum_i n_i^{\text{eq}} = \rho$ ,  $\sum_i n_i^{\text{eq}} \vec{c}_i = \rho \vec{u}$ , and the stress condition  $\sum_i n_i^{\text{eq}} c_{i\alpha} c_{i\beta} = \rho u_\alpha u_\beta + P_{\alpha\beta}$ . Here  $P_{\alpha\beta}$  is the local stress tensor, which takes the form  $P_{\alpha\beta} = \delta_{\alpha\beta} c_s^2 \rho$  in an ideal gas with the isothermal speed of sound  $c_s$ . Other equations of state can be implemented as well.

This summarizes the basic structure of LB algorithms. By means of a Chapman-Enskog equation, one can show that LB fluids reproduce the Navier-Stokes equations on sufficiently large length and time scales and can thus be used as Navier-Stokes solver (Benzi et al. (1992); Chen & Doolen (1998)). Such expansions also give analytical expressions for the shear viscosity as a function of the density and the relaxation time(s). The pseudo equilibrium distribution defines an intrinsic fluid temperature (which enters through the stress condition). Nevertheless, the original formulation of the LB algorithm is fully deterministic. Fluctuations can be incorporated by adding a stochastic term in the collision step, which must to be chosen carefully such that the fluctuation-dissipation relation is fulfilled (Dünweg et al. (2007); Yeomans (2006)).

LB strategies for implementing hydrodynamic boundary conditions are very similar to those used in MPCD. No-slip boundary conditions can be realized by "bounce-back" boundary conditions where the velocities are reversed at the walls. In practice, one introduces a layer of buffer nodes behind the wall, which contains distributions with reversed velocities. This generates a no-slip boundary condition at a hypothetical wall-plane which is half way between the real nodes and the buffer nodes. Generalizations for moving boundaries are available (Ladd (1994a)). Full-slip boundaries can be realized with specular reflection rules. A simple way to generate partial-slip boundary conditions is to blend bounce-back and specular boundary conditions with a weighting factor  $r$ . This generates finite slip lengths which depend on  $r$  in a complex nonlinear manner (Ahmed & Hecht (1999); Succi (2002)). Alternative approaches generate wall slip by implementing conservative wall-fluid interactions which mimic the effect of rough surfaces (Kunert & Harting (2008)) or impose fixed velocities at the boundaries (Hecht & Harting (2010); Zou & He (1997)). The latter approach is particularly interesting in the context of EOF.

Since the LB method is not a particle-based method, coupling LB fluids to nanoparticles is less straightforward than in the case of DPD or MPCD. One popular option is to treat the nanoparticles as extended objects with a well-defined hard surface which interact with the LB fluid *via collisions* at the boundaries. The nanoparticle surface is first mapped on a lattice grid. Then suitable boundary conditions are implemented, which take into account the local velocity of the fluid. From the transfer of momentum during the reflections of the LB fluid, one can evaluate the forces and torques on the nanoparticle and update its velocity and angular velocity accordingly (Ladd (1994a;b)).

An alternative approach which is more appropriate for studying macromolecules is to couple the nanoobjects to the LB fluids *via the forces* that they exert on each other (Ahlrichs & Dünweg (1999)). We will illustrate this method at the example of a single point particle with velocity  $\vec{V}$  and position  $\vec{R}$ . In a fluid flow  $\vec{u}$ , this particle experiences a force

$$\vec{F} = \vec{F}^C - \zeta(\vec{V} - \vec{u}(\vec{R}, t)) + \vec{F}^R. \quad (40)$$

The first term  $\vec{F}^C$  subsumes the conservative forces acting on the particle, the second represents the frictional drag in the LB fluid, and the last one ( $\vec{F}^R$ ) a stochastic random force which satisfies the fluctuation dissipation relation

$$\langle F_\alpha^R(t) F_\beta^R(t') \rangle = 2\zeta k_B T \delta_{\alpha\beta} \delta(t - t'). \quad (41)$$

The dissipative and stochastic terms describe the forces on the particle that stem from the particle-fluid interactions. To ensure momentum conservation (actio=reactio), the opposite forces must be transferred to the fluid. This is done by introducing an additional forcing term in the collision step. Extended nanoobjects are taken to consist of several pointlike monomers. This is clearly a reasonable picture in the case of polymers. The approach can also be used for studying colloids, which are then constructed from point-like interaction sites (raspberry model) (Lobaskin & Dünweg (2004)). In that case, one has the somewhat unphysical situation that the LB fluid may flow through the particle. In practice, however, the dissipative coupling  $\zeta$  ensures that the LB fluid locally moves at exactly the same speed than the colloid – which effectively generates no-slip boundary conditions at the surfaces.

### 3.1.4 Brownian Dynamics simulations (BD)

As discussed in section 2.1, fluid flow is laminar on the nanoscale. Therefore, mesoscale simulation methods must not necessarily be able to reproduce the full Navier-Stokes equations, it suffices if they reproduce the Stokes equation. Nanoscale hydrodynamic flow patterns build up almost instantly, on time scales much smaller than the motion of the nanoparticles: The fluid follows the particles quasi adiabatically. This motivates BD approaches where the solvent is omitted altogether. In BD simulations, the particles undergo overdamped Brownian motion and the effect of hydrodynamics are mimicked by the implementation of suitable hydrodynamic interactions (7). The equations of motion for a system of particles  $i$  subject to the forces  $\vec{F}_i$  in an external flow field  $\vec{u}(\vec{r})$  then have the general form (Ermak & McCammon (1978))

$$\dot{r}_{i\alpha} = u_\alpha(\vec{r}_i) + \sum_j \sum_\beta D_{i\alpha,j\beta} F_{j\beta} + \sum_j \sum_\beta \zeta_{i\alpha,j\beta} \theta_{j\beta}, \quad (42)$$

where  $\mathbf{D}$  is a symmetrical mobility matrix,  $\theta$  is an uncorrelated Gaussian distributed white noise with

$$\langle \theta_{i\alpha} \rangle = 0 \quad \text{and} \quad \langle \theta_{i\alpha}(t) \theta_{j\beta}(t') \rangle = 2 k_B T \delta_{ij} \delta_{\alpha\beta} \delta(t - t'), \quad (43)$$

and the matrix  $\zeta$  is chosen such that  $\zeta \zeta^T = \mathbf{D}$ . The indices  $i, j$  denote particles and  $\alpha, \beta$  cartesian coordinates.

The mobility matrix  $\mathbf{D}$  incorporates the effect of hydrodynamic interactions in the given channel geometry. In free solution, it would seem natural to construct it from the Oseen tensor

$$D_{i\alpha,j\beta} = \delta_{ij} \delta_{\alpha\beta} \mu_i^{(0)} + (1 - \delta_{ij}) T_{\alpha\beta}(\vec{r}_i - \vec{r}_j), \quad (44)$$

where  $\mathbf{T}$  is the Oseen tensor of a point particle given by Eq. (8) and  $\mu_i^{(0)}$  the mobility of a single particle, which can be approximated by the inverse Stokes friction  $\mu^{(0)} = (6\pi\eta a)^{-1}$  for spherical particles of radius  $a$ . Unfortunately, the naive implementation of this expression

leads to numerical problems due to the fact that  $\mathbf{D}$  may become singular. Therefore, the Oseen tensor  $\mathbf{T}(\vec{r})$  in Eq. (44) is usually replaced by the more refined Rotne-Prager tensor

$$T_{\alpha\beta}^{RP}(\vec{r}) = \frac{1}{8\pi\eta r} \left( \delta_{\alpha\beta} + \hat{r}_\alpha \hat{r}_\beta + 2\left(\frac{a}{r}\right)^2 \left( \frac{1}{3} \delta_{\alpha\beta} - \hat{r}_\alpha \hat{r}_\beta \right) \right), \quad (45)$$

which describes approximately the hydrodynamic interactions between spheres of radius  $a$  with no-slip surfaces (Dhont (1996)).

In confined geometry, the hydrodynamic interactions depend on the boundary conditions at the walls. General analytical expressions for  $\mathbf{D}$  are not available. In some cases, however, solutions for  $\mathbf{D}$  can be constructed from the knowledge of the mobility matrix in free solution. For example,  $\mathbf{D}$  can be calculated in slit geometries calculated by a summation over a series of image charges (Kekre et al. (2010)).

Even though BD simulations disregard the solvent degrees of freedom, they quickly become very expensive for large macroion numbers  $n$ . This is not only due to the long-range nature of the hydrodynamic interactions, but also because of the stochastic contribution to Eq. (42), which has to be determined from the mobility matrix. For symmetric and positively definite matrices  $\mathbf{D}$ , a matrix  $\zeta$  satisfying  $\zeta\zeta^T = \mathbf{D}$  can be calculated in a straightforward manner, *e.g.*, *via* a Cholesky decomposition. Unfortunately, the costs for this operation scale with  $n^3$ . This prohibits exact large scale simulations of many-particle systems with BD, and schemes for treating the noise at an approximate level are necessary (Banchio & Brady (2003); Fixman (1986); Geyer & Winter (2009)). The evaluation of the hydrodynamic interactions themselves can be accelerated using methods developed for electrostatics such as fast multipole techniques.

### 3.2 Simulation of electrokinetic phenomena

Simulations of hydrodynamic phenomena are challenging. Simulations of electrohydrodynamic phenomena are even worse due to the long-range nature of the Coulomb interactions. Even though the static equilibrium interactions between macroions are exponentially screened in fluids of high ionic strength, the microions in solution still influence the hydrodynamic flows and can thus not be ignored altogether.

In electrophoresis, we have discussed in Sec. 2.2.3 that the microions also screen hydrodynamic interactions in some sense. Early simulation approaches have therefore resorted to local BD simulation models where both hydrodynamic and electrostatic interactions are neglected (Viovy (2000)) (see Sec. 3.2.2). Such models can be quite successful. Nevertheless, they clearly disregard important physics. For example, hydrodynamic interactions are "screened" only with respect to the response of particles to slowly varying external electric fields, but not with respect to other external forces, internal forces, or diffusion. Moreover, time-dependent phenomena where ion relaxation becomes important cannot be studied within such simplified BD schemes.

To go beyond local BD models, one must account for both hydrodynamics and electrostatics simultaneously. Three types of approaches have been pursued in the past: Fully explicit models with explicit solvent and explicit microions, fully implicit models with implicit solvent and implicit microions, and mixed schemes where either the solvent or the microions are

treated at an implicit level. In the following, we shall give a brief overview over these approaches with a few examples. More detailed reviews can be found in (Pagonabarraga et al. (2010); Slater et al. (2009)).

### 3.2.1 Fully explicit models

In fully explicit simulation approaches, the fluid is treated by explicit mesoscale methods such as DPD, MPCD, or LB, and the charges are represented by explicit charged point particles which are coupled to the fluid following the prescriptions described in the Sections 3.1.1–3.1.3. This is computationally expensive, but feasible. To give an example, the electrophoresis of short polyelectrolytes in free solutions has been studied recently both by fully explicit LB and MPCD simulations (Grass et al. (2008); Frank & Winkler (2008)). The computationally most challenging part of the simulations is the evaluation of the electrostatic interactions, especially in systems with high salt concentrations. Large efforts are currently dedicated to developing efficient methods for treating Coulomb interactions in many-particle systems (Sutmann (2009)), which will undoubtedly also benefit electrokinetic simulations in the future.

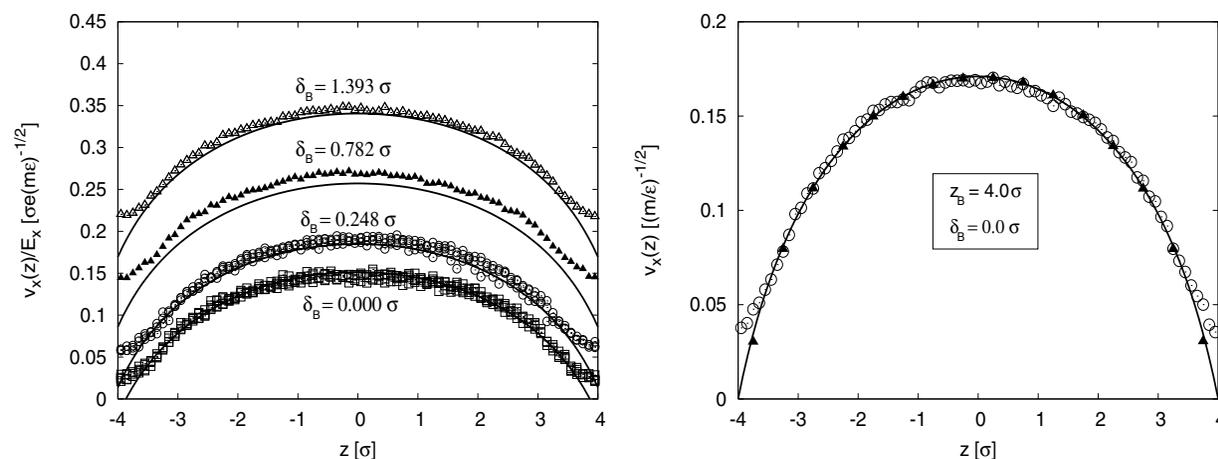


Fig. 2. **Left side:** Flow profiles for the DPD-method for various slip lengths in the Poisson-Boltzmann regime. The hydrodynamic boundary positions for the DPD-method are at  $|z_B| = (3.866 \pm 0.265)\sigma$ . The straight lines represent the theoretical prediction of Eqn.(20). **Right side:** Flow profile for the DPD-(circles) and the LB-method (triangles) with no-slip boundary conditions and  $E_x = 1.0\epsilon/e\sigma$ . The straight line is the theoretical prediction of Eqn. (20) with  $|z_B| = 4.0\sigma$  and  $\delta_B = 0.0\sigma$ . From Ref. (Smiatek et al. (2009)).

Here, we will illustrate the fully explicit approach at the example of two applications in microchannels. The first one is a comparative study of counterion-induced EOF in symmetric planar slit channels using fully explicit DPD and LB simulations (Smiatek et al. (2009)). The theoretical prediction for the EOF profiles in this case is given by Eq. (20). We have studied such flows by two complementary approaches, DPD and LB, and devised schemes how the model parameters could exactly be mapped onto each other. Fig. 2 demonstrates that the results of DPD and LB simulations are in excellent agreement with each other and with theory. The left graph illustrates remarkable flow enhancement due to slippage already in the presence of small surface slip. The examples shown in Fig. (2) refer to situations where the Poisson-Boltzmann theory is applicable, *i.e.*, the microions are monovalent and the charges on the channel surfaces are moderate. With fully explicit simulations, one can also study regimes

where the mean-field assumption breaks down. In that case, explicit simulations show that the relation between flow and electrostatic potential, Eq. (19), remains valid even though the ion distribution can no longer be calculated analytically (Smiatek et al. (2009)).

The electrophoretic response of polyelectrolytes in symmetric slit channels is considered in our second example. As discussed in Section 2.2.3, it results from the combination of bare electrophoresis and background EOF. From Eqs. (21), (23) and (25), one can derive the relation

$$\frac{\mu_t}{\mu_{EOF}} = 1 + \frac{\mu_e}{\mu_{EOF}^0(1 + \delta_B \kappa_s)}. \quad (46)$$

This relation is verified in Fig. 3 (left). We have studied the electrophoresis of a charged polyelectrolyte in several electrolytes with different salt concentrations, confined by charged planar walls with different slip lengths. The behavior of the electrophoretic mobility is in very good agreement with the theoretical prediction. Most notably, the EOF flow may reverse the sign of the electrophoretic mobility, such that the polyelectrolyte migrates *against* the applied field. Following the discussion in Section 2.2.3, we expect the bare mobility of the polyelectrolytes to be independent of chain length in solvents of high ionic strength. Fig. 3 (right) demonstrates that this is indeed the case.

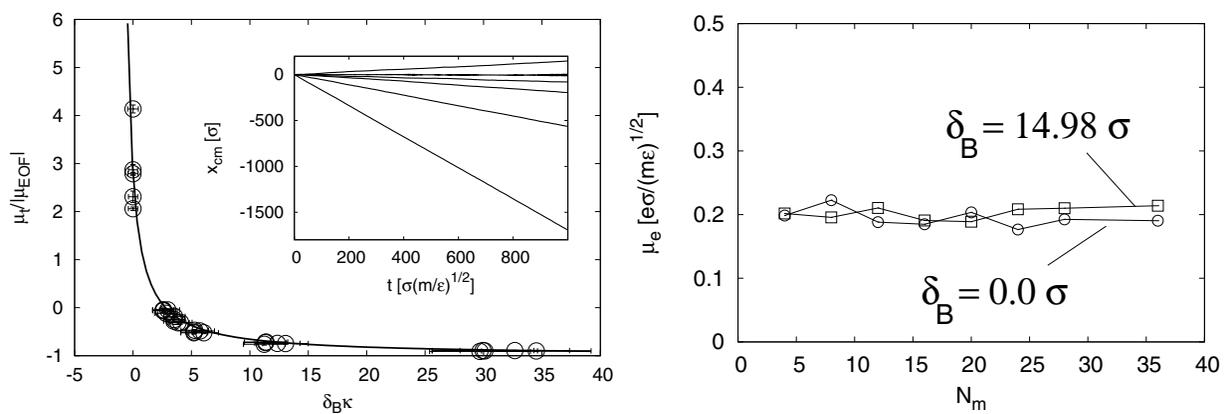


Fig. 3. **Left side:** Ratio  $\mu_t/|\mu_{EOF}|$  plotted against  $\delta_B \kappa_s$  for chains of length  $N_m = 20$  and several slip lengths and salt concentrations. The black line is the theoretical prediction of Eqn. (46). The ratio  $\mu_e/\mu_{EOF}^0$  has been fitted to  $-3.778 \pm 0.128$ . Negative values of  $\mu_t/|\mu_{EOF}|$  indicate negative total mobilities of the polyelectrolyte. **Inset:** Total displacement of the polyelectrolyte's center of mass for different boundary conditions and a salt concentration of  $\rho_s = 0.05625\sigma^{-3}$ , in units of the ion diameter  $\sigma$ . The total mobility becomes negative for  $|\mu_e| \ll |\mu_{EOF}|$ . The lines correspond from top to bottom to the slip lengths  $\delta_B \approx (0.00, 1.292, 1.765, 2.626, 5.664, 14.98)\sigma$ . Thus larger slip lengths indirectly enhance the total mobility of the polyelectrolyte. **Right side:** Bare electrophoretic mobility for different monomer lengths  $N_m$  in confined geometry in the presence of salt for two slip lengths.

Finally in this section, we mention a special boundary effect which has not been discussed so far and is often neglected in electrokinetic simulations: The dielectric contrast between water and typical microchannel materials is very high. Therefore, the surfaces are polarized and charges are induced at the boundaries. Tyagi et al. have recently developed a general method that allows one to account for this effect (Tyagi et al. (2007; 2010)). First preliminary studies

seem to indicate that it does not influence the transport properties very much on length scales larger than the Debye length.

### 3.2.2 Fully implicit models

In fully implicit approaches, both solvent and microion degrees of freedom are disregarded and their effect on the particles is incorporated in effective interactions and an effective mobility matrix (see Eq. (42)). This implies an adiabatic approximation: Both the hydrodynamic flows and the local microion distributions are assumed to adjust instantaneously to the macroion configuration. Whereas this approximation can be justified in the case of flows at low Reynolds numbers (as discussed in Section 3.1.4), it is somewhat questionable in the case of the ions. We have discussed important phenomena in Section 2.2 which cannot be captured within such an approach, *e.g.*, the relaxation effect in electrophoresis. Relaxation processes in the ionic double layer of charged colloids have characteristic time scales ranging from microseconds (Maxwell-Wagner relaxation, ionic migration sets in) to 10 milliseconds (the thickness of the Debye layer adjusts), which is close to the characteristic time scales of many mesoscale simulations. Thus fully implicit methods are hardly suited for studying dynamical processes with a time evolution. They can however be used to study quasi-stationary situations such as electrophoresis or pressure-driven transport at a semi-quantitative level.

A second drawback of implicit schemes is that they usually rely on the Debye-Hückel approximation, which has a limited range of validity as discussed in Section 2.2.1. Nevertheless, it can often be applied successfully even in the nonlinear regime as long as one accounts for nonlinear effects by appropriate charge renormalization or renormalization of the local Debye Hückel screening length (see 2.2.1).

Fully implicit simulations are stochastic simulations with the structure of BD simulations (Eq. (42)). Setting up BD schemes for nanoparticles in external electric fields is slightly subtle due to the fact that one has two mobility matrices: One ( $\mathbf{D}^C$ ) which describes the hydrodynamic interactions in response to conservative forces and is related to the unscreened Greens function Eq. (8) or (45) in free solution, and one ( $\mathbf{D}^E$ ) which describes the hydrodynamic interactions in response to external electric fields, and can be related to the screened Greens function (29) in free solution if the external fields vary slowly compared to the Debye Hückel screening length. A clean way of dealing with this situation is to treat both types of responses on separate footings: In a constant or slowly varying electric field  $\vec{E}$  and external flow field  $\vec{u}(\vec{r})$ , the equation of motion of a system of particles  $i$  with charges  $q_i$  (the equivalent of Eq. (42)) is then given by (Kekre et al. (2010))

$$\dot{r}_{i\alpha} = u_\alpha(\vec{r}_i) + \sum_j \sum_\beta D_{i\alpha,j\beta}^C F_{j\beta}^C + \sum_j \sum_\beta \xi_{i\alpha,j\beta} \theta_{j\beta} + \sum_j \sum_\beta D_{i\alpha,j\beta}^E q_j E_\beta, \quad (47)$$

where  $\mathbf{D}^C$  and  $\mathbf{D}^E$  are given by

$$D_{i\alpha,j\beta}^C = \delta_{ij} \delta_{\alpha\beta} \mu_i^{(0)} + (1 - \delta_{ij}) T_{\alpha\beta}(\vec{r}_i - \vec{r}_j)$$

$$D_{i\alpha,j\beta}^E = \delta_{ij} \delta_{\alpha\beta} \mu_i^{\text{el}} + (1 - \delta_{ij}) G_{\alpha\beta}(\vec{r}_i - \vec{r}_j)$$

and the matrix  $\zeta$  meets the requirement  $\zeta\zeta^T = \mathbf{D}^C$ . Here  $\mu_i^{(0)}$  is the bare frictional mobility of the particle  $i$ ,  $\mu_i^{\text{el}}$  its electrophoretic mobility, and  $\mathbf{T}$ ,  $\mathbf{G}$  are the screened and unscreened hydrodynamic Greens functions, which can be approximated by the Rotne-Prager tensor (45) and the screened Greens function (29) (or the corresponding Rotne-Prager version (Fischer et al. (2008))) in free solution. It is important to note that  $\vec{F}_j^C$  subsumes all effective conservative interactions between ions, including those of electrostatic origin. One can show rigorously that a procedure where (microion) degrees of freedom are integrated out adiabatically only renormalizes the conservative potentials of the remaining macroions and not their mobility coefficients. Thus, the Coulomb potentials are replaced by screened Yukawa potentials, but the hydrodynamic interactions remain long-range. The situation is different in electrophoresis since this is an inherently nonequilibrium process.

Solving Eq. (47) for many-particle systems is expensive due to the long-range hydrodynamic interactions and the noise issue discussed in Section 3.1.4. If the dominating forces in a system are due to external electric fields, it may become acceptable to neglect the long-range hydrodynamic contributions. In nanochannels, this approximation can further be justified by the fact that hydrodynamic interactions are screened in confinement (see Section 2.1.2). For these reasons, simulation approaches have been popular already early on where both hydrodynamics and electrostatics were neglected entirely. The motion of macroions in an external electric field  $\vec{E}(\vec{r})$  and flow field  $\vec{u}(\vec{r})$  is then described by simple local Langevin dynamics, where the force  $\vec{F}_i$  on a particle  $i$  at position  $\vec{r}_i$  is given by the sum

$$\vec{F}_i = \vec{F}_i^C + \vec{F}_i^D + \vec{F}_i^R \quad (48)$$

of conservative contributions  $\vec{F}_i^C = q_i\vec{E}(\vec{r}_i) + \vec{F}_{i,\text{int}}^C$ , the dissipative drag force  $\vec{F}_i^D = -\zeta(\vec{v}_i - \vec{u}(\vec{r}_i))$  and a Gaussian distributed random force  $\vec{F}_i^R$  which satisfies the fluctuation dissipation relation  $\langle F_{i\alpha}^R(t) F_{j\beta}^R(t') \rangle = 2\zeta k_B T \delta_{ij} \delta_{\alpha\beta} \delta(t - t')$ . The internal contributions  $\vec{F}_{i,\text{int}}^C$  subsume the screened electrostatic interactions between particles as well as other forces with non-electrostatic origin.

The situation is particularly simple if the external flow  $\vec{u}(\vec{r})$  has been generated by EOF and the channel has the similitude property (Section 2.2.2), *i.e.*,  $\vec{u}(\vec{r}) = \mu_{\text{EOF}}\vec{E}(\vec{r})$ . In that case, the total force on the particle  $i$  is given by

$$\vec{F}_i = \vec{F}_{i,\text{int}}^C + (q_i + \mu_{\text{EOF}}\zeta)\vec{E}(\vec{r}_i) - \zeta\vec{v}_i + \vec{F}_i^R. \quad (49)$$

Thus the EOF flow  $\vec{u}(\vec{r})$  simply has the effect of renormalizing the charge  $q_i$  (Streek et al. (2005)).

Simple local BD approaches such as Eq. (49) have been used quite successfully to study electrophoresis in complex geometries, and they can yield good qualitative and even semiquantitative agreement with experiments (Streek et al. (2004); Viovy (2000)). As an example, Fig. 4 shows results from such a BD simulation of DNA electrophoresis in a structured microchannel. Both in experiments and simulations, one observes a two-state behavior where the DNA switches between a "slow" and "fast" migration mode. This is reflected by an inhomogeneous average monomer distribution in the channel (Streek et al. (2005)).

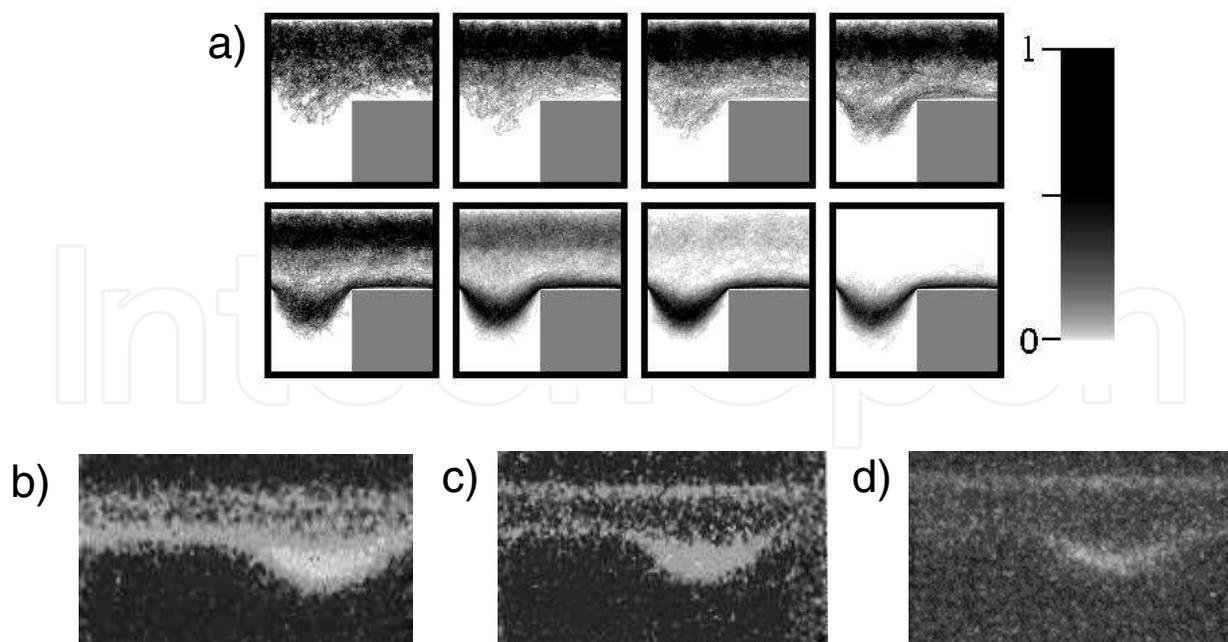


Fig. 4. **Top panel** (a): Monomer density histograms obtained from local BD simulation of electrophoresis of charged chains with different lengths  $N = 50, 100, 120, 140$  (top, left to right) and  $N = 170, 200, 250$  and  $320$  (bottom) at. **Bottom panel**: Monomer density histograms obtained from experiment at applied voltages (b)  $E = 43V/cm$ , (c)  $E = 86V/cm$ , and (d)  $E = 130V/cm$  for  $\lambda$ -DNA in  $5\mu m$  constrictions. The fluorescence intensity decreases because the illumination time was kept constant. Note the depletion zone in the center of the channel. From (Streek et al. (2005))

Despite these successes, the complete neglect of hydrodynamic interactions sometimes simplifies the situation too much even in simulations of electrophoretic transport. In these cases, one may still choose to work with the screened hydrodynamic Greens function  $\mathbf{G}$  for reasons of efficiency and neglect the unscreened contributions. Löwen et al have demonstrated that such a scheme can be used successfully to describe the effect of hydrodynamic interactions in systems of oppositely charged colloids subject to an external field. Even the screened hydrodynamic interactions have a noticeable effect on structure formation and "laning" in such systems (Rex & Löwen (2008)).

### 3.2.3 Mixed schemes

Mixed schemes for electrohydrodynamic simulations treat either the solvent or the microions at an implicit level.

In the first approach, both microions and macroions are represented by explicit particles which interact with each other by means of the full Rotne-Prager tensor, Eq. (45) (Fischer et al. (2008); Kim & Netz (2006)). These methods are motivated by the fact mentioned earlier that hydrodynamic flows adjust very rapidly, whereas ion rearrangement processes can be slow on mesoscopic time scales. Unfortunately, they suffer from the bad scaling properties of BD simulations and have therefore not been used to study very large systems.

In the second type of approach, the solvent is modeled as an explicit fluid, but microions are no longer represented by individual particles. Many kinetic processes in electrolyte solutions

do not depend on ion correlations and can thus be described satisfactorily at the mean-field level. To access larger length scales in simulations, one may therefore resort to methods that treat the ions at the level of the Nernst-Planck equation (12). So far, efforts to develop such methods have mainly focused on LB approaches (Capuani et al. (2004); Horbach & Frenkel (2001); Pagonabarraga et al. (2005); Warren (1997)).

A number of authors have proposed simplified schemes that capture basic electrokinetic phenomena at a heuristic level in explicit fluid simulations. For example, one simple way to deal with EOF is to impose a velocity boundary condition (Eq. (23)) at the walls of a microfluidic channel (Duong Hong et al. (2008)). Similarly, the electrohydrodynamic screening effect in electrophoresis can be mimicked in a LB simulation by adding a slip term to the Stokes drag on the particle in the LB flow. The dissipative contribution  $F^D = -\zeta(\vec{V} - \vec{u}(\vec{R}, t))$  to Eq. (40) is then replaced by a modified drag force (Hickey et al. (2010))

$$F^D = -\zeta(\vec{V} - \vec{u}(\vec{R}, t) - \mu_0 \vec{E}) \quad (50)$$

Unfortunately, this elegant idea cannot easily be transferred to DPD and MPCD simulations. Duong-Hong et al. have proposed an alternative approach where charges are assigned to DPD or MPCD particles that are within a certain distance (the Debye length) of the macroions or charged walls. These particles are then dragged along by the electric field, which can be used to mimic electrophoretic retardation, electrohydrodynamic screening effects, and EOF generation on channel surfaces in a unified manner (Duong Hong et al. (2008b)).

#### 4. Summary and outlook

With the advent of faster computers and improved simulation methods for electrostatics and hydrodynamics, electrohydrodynamic simulations become an increasingly attractive tool for studying and predicting electrokinetic phenomena in microgeometries. In this chapter we have introduced the main concepts behind coarse-grained mesoscale simulations of micro- and nanofluidic flow and transport. We have focused on boundary effects and mean-field phenomena, *e.g.*, phenomena that can be rationalized within the Poisson-Boltzmann theory. Several important phenomena such as counterion condensation, overcharging, shape transformations of the polyelectrolytes, confinement effects, to name a few, have not been discussed in this work. Also, important problems of current interest such as electrophoresis with obstacle collisions or polymer nanopore translocations, have not been mentioned. For further reading, we refer the interested reader to the reviews (Boroudjerdi et al. (2005); Graham (2011); Pagonabarraga et al. (2010); Slater et al. (2009); Viovy (2000)).

We have shown that mesoscopic simulation methods are powerful tools to explore fluid behavior on the micro- and nanoscale. Such simulations not only help to understand and optimize transport processes in micro- and nanofluidic channels, they can also be used to study basic mechanisms for many physical processes of life, since such processes often take place in confinement (*e.g.*, cells) and in aqueous solutions with high ion concentrations. We expect that the mesoscale techniques introduced in this chapter, combined with microscopic and macroscopic approaches, will open new routes towards a better understanding of transport phenomena on the micro- and nanoscale.

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(<http://www.icp.uni-stuttgart.de/~icp/ESPResSo>)

(Arnold et al. (2006)). We thank the Volkswagen Stiftung and the Deutsche Forschungsgemeinschaft for funding and the HLR Stuttgart, NIC Jülich and Arminius PC<sup>2</sup>-cluster in Paderborn for computing times.

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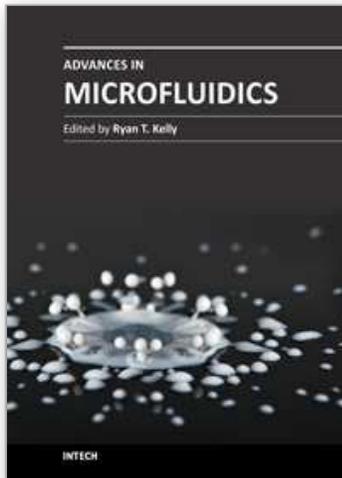
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Phone: +86-21-62489820  
Fax: +86-21-62489821

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